PRERNA NEET TEST SERIES 2025

Syllabus

PHYSICS : Complete Syllabus; **CHEMISTRY :** Complete Syllabus **BOTANY :** Complete Syllabus; **ZOOLOGY :** Complete Syllabus

INSTRUCTIONS

- 1. Immediately fill in the particulars on this page of the Test Booklet with Blue/Black Ball Point Pen only.
- 2. The test is of 3 hours 20 minutes duration and the test booklet contains 200 Multiple choice questions. Which have four options with a single correct answer.
- 3. This test consists of Physics, Chemistry, Botany and Zoology questions with equal weightage of 180 marks.
- 4. Each question is of 4 marks. For each correct response the candidate will get 4 marks. For each incorrect response,
 1 mark will be deducted from the total score. The maximum marks are 720.
- 5. There are four parts in the question paper, consisting Part-I Physics (Q.no.1 to 50), Part-II Chemistry (Q.no. 51 to 100), Part-III Botany (Q. no. 101 to 150) and Part-IV Zoology (Q. no.151 to 200). Each part is divided into two Sections, Section-A consists of 35 multiple choice questions & Section-B consists of 15 Multiple choice questions, out of these 15 questions candidates can choose to attempt any 10 questions.

	Parts Sections	Physics	Chemistry	Botany	Zoology	Тс	otal
Questions	Section A	35	35	35	35	140	200
	Section B	15	15	15	15	60	200
To Attempt	Section A	35	35	35	35	140	180
	Section B	10	10	10	10	40	160

- 6. Candidates are advised to read all 15 questions in each subject of Section-B before they start attempting the question paper. In the event of a candidate attempting more than ten questions, the first ten questions answered by the candidate shall be evaluated.
- 7. Use Blue/Black Ball Point Pen only for writing particulars/marking responses on OMR Sheet.
- Do not fold or make any stray marks on the Answer Sheet. Rough work is to be done on the space provided for this purpose.



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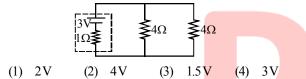
FT-01

PART-I: PHYSICS

Section-A

- 1. If the temperature of a gas is increased from 27°C to 159°C, then the percentage increase in the rms speed of the gas molecules is [NCERT-XI, Page 250] (1) 5 (2) 10 (3) 15 (4) 20
- 2. In a Young's double slit experiment, if the wavelength of light is increased by 50% and the distance between the slits is doubled then the percentage change in fringe width [NCERT-XII, Page 266] is (1)15

- 3. A vernier callipers has 20 divisions on the vernier scale, which coincides with 19th division on the main scale. The least count of the instrument is 0.1 mm. One main scale division is equal to (1) 1mm (2) 0.5 mm (3) 2 mm (4) 5mm
- 4. In the given circuit, the terminal potential difference of the cell is : [NCERT-XII, Page 94]



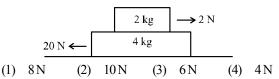
The magnetic moments associated with two closely 5. wound circular coils A and B of radius $r_A = 10$ cm and $r_B = 20$ cm respectively are equal if: (Where N_A, I_A and N_B, I_{B} are number of turn and current of A and B respectively) [NCERT-XII, Page 128]

(1)
$$2N_{A}I_{A} = N_{B}I_{B}$$
 (2) $N_{A} = 2N_{B}$
(3) $N_{A}I_{A} = 4N_{B}I_{B}$ (4) $4N_{A}I_{A} = N_{B}I_{B}$

In the arrangement shown in the figure, the coefficient of 6. friction between two blocks is 0.5. The force of friction between the two blocks is (Assume that the 4 kg block is placed on a smooth horizontal surface.)

(Acceleration due to gravity = 10 ms^{-2} .)

[NCERT-XI, Page 60, 61]



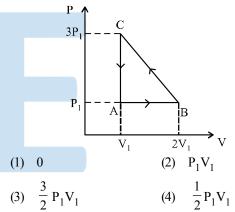
Primary side of a transformer is connected to 230 V, 50 Hz 7. supply. Turns ratio of primary to secondary winding is 10: 1. Load resistance connected to secondary side is 46Ω. The power consumed in it is : [NCERT-XII, Page 195] (1) 12.5 W (2) 10.0 W

8. Given below are two statements: Statement I : Most of the mass of the atom and all its positive charge are concentrated in a tiny nucleus and the electrons revolve around it, is Rutherford's model.

Statement II : An atom is a spherical cloud of positive charges with electrons embedded in it, is a special case of Rutherford's model. [NCERT-XII, Page 293] In the light of the above statements, choose the most appropriate from the options given below

- (1) Statement I is true but Statement II is false
- Both Statement I and Statement II are true (2)
- (3)Both Statement I and Statement II are false
- (4)Statement I is false but Statement II is true
- 9. Two long parallel straight metal wires A and B carrying currents 12 A and 36 A respectively, in the same direction are separated by 50 cm. The point relative to A, where the resultant magnetic induction between the two wires due to the currents is zero, will be [NCERT-XII, Page 123] (1) 90 cm (2) 7.5 cm (3) 28 cm(4) $12.5 \,\mathrm{cm}$
- 10. The net work done by an ideal gas going through the cycle as shown in the P - V diagram below is

[NCERT-XI, Page 235]



An electric charge $10^{-6} \mu C$ is placed at origin (0, 0) m of 11. X-Y co-ordinate system. Two points P and Q are situated at $(\sqrt{3}, \sqrt{3})$ m and $(\sqrt{6}, 0)$ m respectively. The potential difference between the points P and Q will be :

[NCERT-XII, Page 48]

(1)
$$\sqrt{3}V$$
 (2) $\sqrt{6}V$ (3) 0 V (4) 3V

- A string of length 1 m and mass 490 g is put under a 12. tension of 25 N. A wave of frequency 120 Hz is sent along it. The speed of this wave is [NCERT-XI, Page 285]
 - (1) $7.14 \,\mathrm{ms}^{-1}$ (2) $0.71 \,\mathrm{ms}^{-1}$ (3) $0.51 \,\mathrm{ms}^{-1}$ (4) $51.0 \,\mathrm{ms}^{-1}$
- 13. A stone of mass 2 kg is tied at one end of a string of length 2 m and is whirled in a horizontal circle. If the string can withstand a maximum tension of 64 N, then the permissible maximum number of rotations per minute is

[NCERT-XI, Page 63]

(1) 19 (2)
$$\frac{60}{\pi}$$
 (3) $\frac{152}{3}\pi$ (4) $\frac{120}{\pi}$

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- 14. Which of the following phenomena does not explain by wave nature of light. [NCERT-XII, Page 280, 281]
 - (A) reflection
 - (B) diffraction
 - (C) photoelectric effect
 - (D) interference
 - (E) polarization

Choose the most appropriate answer from the options given below :

- (1) E only
- (2) C only

(1)

- (3) B, D only
- (4) A, C only

15. A body of mass 2 kg begins to move under the action of a

time dependent force given by $\vec{F} = (6t\hat{i} + 6t^2\hat{j})N$. The

power developed by the force at the time t is given by:

		[NCERT-XI, Page 83]
$(6t^4+9t^5)W$	(2)	$(3t^3 + 6t^5)$ W

- (3) $(9t^5 + 6t^3)$ W (4) $(9t^3 + 6t^5)$ W
- 16. A light ray incidents normally on one surface of an equilateral prism. The angle of deviation of the light ray is (refractive index of the material of the prism = $\sqrt{2}$)

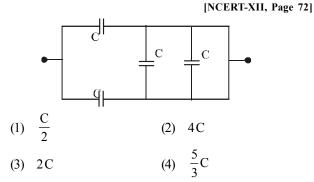
[NCERT-XII, Page 239, 240]

(1)	60°	(2)	30°
(3)	0°	(4)	120°

Match the physical quantities of Column I with their 17. mathematical expressions in Column II.

[NCERT-XII, Page 112, 116, 123, 126] Column I Column II $\mu_0 i$ (A) Torque on a circular (p) 2Rcurrent loop placed in uniform magnetic field (B) Force per unit length (q) $iAB\sin\theta$ between parallel current carrying wires тV (C) Magnetic field at the (r) qBcentre of a circular current carrying loop. $\mu_0 \, l_1 \, l_2$ (D) Radius of circular path (s) of a charge particle moving in uniform magnetic field. (1) (A) \rightarrow (q); (B) \rightarrow (p); (C) \rightarrow (r); (D) \rightarrow (s) (2) (A) \rightarrow (q); (B) \rightarrow (q); (C) \rightarrow (s); (D) \rightarrow (r) (3) (A) \rightarrow (s); (B) \rightarrow (r); (C) \rightarrow (q); (D) \rightarrow (p)

- (4) (A) \rightarrow (q); (B) \rightarrow (s); (C) \rightarrow (p); (D) \rightarrow (r)



The equivalent capacitance of the combination shown is

19. Which of the following pairs has same dimensions?

[NCERT-XI, Page 7]

- (1) Current density and charge density
- (2)Angular momentum and linear momentum
- Spring constant and surface energy (3)
- Force and torque (4)

18.

20. A metal wire with circular cross section and length one metre is pulled with tensile force of 1000 N on each side. For the wire to be stretched not more than 0.25 cm, the minimum diameter of the wire required is

(Young's modulus of the metal = 10^{11} Pa,

take $\sqrt{\pi} = 1.77$)	[NCERT-XI, Page 170]
(1) 1.13 mm	(2) 2.26 mm
(3) 4.12 mm	(4) 3.1 mm

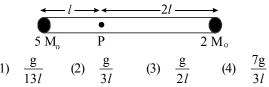
21. Given below are two statements : one is labelled a Assertion (A) and the other is labelled as Reason(R)

Assertion (A) : Work done by electric field on moving a positive charge on an equipotential surface is always zero. **Reason (R) :** Electric lines of forces are always perpendicular to equipotential surfaces.

In the light of the above statements, choose the most appropriate answer from the options given below :

[NCERT-XII, Page 54]

- Both (A) and (R) are correct but (R) is not the correct (1)explanation of (A)
- (2) (A) is correct but (R) is not correct
- (3) (A) is not correct but (R) is correct
- (4) Both (A) and (R) are correct and (R) is the correct explanation of (A).
- 22. A rigid massless rod of length 3*l* has two masses attached at each end as shown in the figure. The rod is pivoted at point P on the horizontal axis (see figure). When released from initial horizontal position, its instantaneous angular acceleration will be: [NCERT-XI, Page 119]



4

23. A uniform solid sphere of radius R produces a gravitational acceleration of a_0 on its surface. The distance of the point from the centre of the sphere where the gravitational

acceleration becomes $\frac{a_0}{4}$ is [NCERT-XI, Page 130]

(1) 4R (2)
$$\frac{3}{2}$$
R (3) 2R (4)

- 24. A light whose electric field vectors are completely removed by using a good Polaroid, allowed to incident on the surface of the prism at Brewster's angle. Choose the most suitable option for the phenomenon related to the prism.
 - (1) Reflected and refracted rays will be perpendicular to each other
 - (2) Wave will propagate along the surface of prism
 - (3) No refraction, and there will be total reflection of light.
 - (4) No reflection and there will be total transmission of light.
- 25. A metal ball of mass 2 kg moving with a velocity of 36 km/h has a head on collision with a stationary ball of mass 3 kg. After the collision, if both balls move together, then the loss in kinetic energy due to collision is [NCERT-XI, Page 84]
 (1) 40 J
 (2) 60 J
 (3) 100 J
 (4) 140 J
- 26. Binding energy of a certain nucleus is 18×10^8 J. How much is the difference between total mass of all the nucleons and nuclear mass of the given nucleus:

			[NCERT-XII, Page 311]
(1)	0.2 µg	(2)	20 μg
(3)	2 µg	(4)	10 μg

27. The equation of motion of a particle executing simple

harmonic motion is given by $x = 3 \sin \left(6t + \frac{\pi}{6} \right)$, where x is

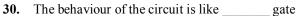
in metres and t is in seconds. The ratio of the potential and kinetic energies of the particle at time t = 0 is

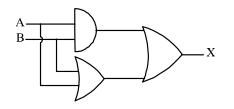
[NCERT-XI, Page 268]

3R

- **28.** If \vec{E} and \vec{B} are the electric and magnetic field vectors of an electromagnetic wave, then the direction of propagation of the electromagnetic wave is
 - (1) along the direction of \vec{E} [NCERT-XII, Page 206]
 - (2) along the direction of \vec{B}
 - (3) parallel to the direction of $\vec{E} \times \vec{B}$
 - (4) perpendicular to the direction of $\vec{E} \times \vec{B}$
- **29.** A water drop breaks into 64 identical droplets of each surface area 10^{-7} m². If the surface tension of water is 0.07 Nm⁻¹, the increase in the surface energy in the process is [NCERT-XI, Page 194]

- (1) 158×10⁻⁹ J
- (2) 432×10⁻⁹ J
- (3) $216 \times 10^{-9} \text{ J}$
- (4) $336 \times 10^{-9} \text{ J}$





(1) OR (2) NOR (3) NAND (4) AND

- A body rotating with uniform acceleration about its geometrical axis makes 8 rotations in the first 2 seconds. The number of rotations the body makes in the next 3 seconds is [NCERT-XI, Page 117] (Initially the body is at rest)
 - (1) 50 (2) 25 (3) 42 (4) 21
- **32. Statement-I**: By doping silicon semiconductor with pentavalent material, the electrons density increases. **Statement-II**: The n-type semiconductor has net negative charge.

In the light of the above statements, choose the most appropriate answer from the options given below :

- [NCERT-XII, Page 330]
- (1) Statement-I is true but Statement-II is false.
- (2) Statement-I is false but Statement-II is true.
- (3) Both Statement-I and Statement-II are true.
- (4) Both Statement-I and Statement-II are false.
- **33.** An electric cable of copper has just one wire of radius 9 mm. Its resistance is 14Ω . If this single copper wire of the cable is replaced by seven identical well insulated copper wires each of radius 3 mm connected in parallel, then the new resistance of the combination will be:

[NCERT-XII, Page 84] Ω (4) 126 Ω

(1) 9Ω (2) 18Ω

34.

Two coils are placed close to each other. The mutual inductance of the pair of coils depends upon

(3) 28Ω

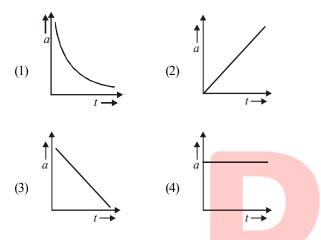
- I. relative position and orientation of the two coils
- II. the materials of the wires of the coils
- III. the rates at which currents are changing in the two coils
- IV. Number of turns in the coils
- Which of the above statements is/are correct?
 - [NCERT-XII, Page 166, 167]
- (1) I and IV(2) II only(3) I and III(4) II and III
- 35. If the angles of projection of a projectile with same initial velocity exceed or fall short of 45° by equal amounts, then the ratio of horizontal ranges is [NCERT-XI, Page 40]
 (1) 1:2
 (2) 1:3
 (3) 1:4
 (4) 1:1

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Section-B

- **36.** The displacement current through the plates of a parallel plate capacitor of capacitance 30 μ F is 150 μ A. The capacitor is charged by a source of varying potential at the rate of [NCERT-XII, Page 203, 204] (1) 3.5 Vs⁻¹ (2) 2 Vs⁻¹ (3) 5 Vs⁻¹ (4) 3 Vs⁻¹
- **37.** The distance travelled by a body moving along a line in time *t* is proportional to t^3 .

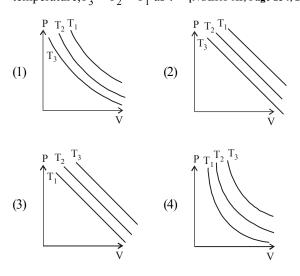
The acceleration-time (*a*, *t*) graph for the motion of the body will be [NCERT-XI, Page 16]



38. The magnifying power of a telescope with tube length 60 cm is 5. Then the focal length of its eye piece is

[NCER<mark>T-XII,</mark> Page 244, 245]

- (1) 20 cm (2) 40 cm (3) 30 cm (4) 10 cm
- 39. Two electric bulbs rated 25W-220 V and 100W-220V are connected in series to a 440 V supply. Which of the bulbs will fuse? [NCERT-XII, Page 93]
 (1) Both
 (2) 100 W
 (3) 25 W
 (4) Neither
- **40.** In an Isothermal change, the change in pressure and volume of a gas can be represented for three different temperature; $T_3 > T_2 > T_1$ as : [NCERT-XI, Page 234, 235]



- 5
- 41. Two parallel large thin metal sheets have equal surface densities 26.4×10^{-12} C/m² of opposite signs. The electric field between these sheets is-(1) 1.5 N/C (2) 1.5×10^{-16} N/C (3) 3×10^{-10} N/C (4) 3 N/C
- **42.** A simple pendulum of length 1 m is freely suspended from the ceiling of an elevator. The time period of small oscillations as the elevator moves up with an acceleration of 2 m/s² is (use g = 10 m/s²) [NCERT-XI, Page 271]

(1)
$$\frac{\pi}{\sqrt{5}}$$
 s (2) $\sqrt{\frac{2}{5}}\pi$ s
(3) $\frac{\pi}{\sqrt{2}}$ s (4) $\frac{\pi}{\sqrt{3}}$ s

- **43.** Energy required to move a body of mass *m* from an orbit of radius 2*R* to 3*R* is [NCERT-XI, Page 134]
 - (1) $GMm/12R^2$ (2) $GMm/3R^2$
 - (3) GMm/8R (4) GMm/6R.
- 44. Match the following. Column II gives $\lambda_{max}/\lambda_{min}$ for the spectral series in column I

Colum	n I		Column II
Lyman	series	(p)	16/7
Balmer	series	(q)	9/5
Pascher	n series	(r)	25/9
Bracket	t series	(s)	4/3
$(A) \rightarrow (A)$	$(q); (B) \rightarrow (r); (B) $	$(C) \rightarrow$	$(\mathbf{p}); (\mathbf{D}) \rightarrow (\mathbf{s})$
$(A) \rightarrow (A)$	(s); (B) \rightarrow (q);	$(C) \rightarrow$	$(\mathbf{p}); (\mathbf{D}) \rightarrow (\mathbf{r})$
$(A) \rightarrow (A)$	$(\mathbf{r}); (\mathbf{B}) \to (\mathbf{q}); (\mathbf{a}) \to (\mathbf{q}); (\mathbf{q}); (\mathbf{q}) \to (\mathbf{q}); (\mathbf{q}); (\mathbf{q}) \to (\mathbf{q}$	$(C) \rightarrow$	$(s); (D) \rightarrow (p)$
	Lyman Balmer Pascher Bracker $(A) \rightarrow (A)$	$(A) \rightarrow (s); (B) \rightarrow (q);$	Lyman series(p)Balmer series(q)Paschen series(r)

- (4) (A) \rightarrow (q); (B) \rightarrow (s); (C) \rightarrow (r); (D) \rightarrow (p)
- **45.** A steel rod with $Y = 2.0 \times 10^{11} \text{ Nm}^{-2}$ and $\alpha = 10^{-5} \text{ °C}^{-1}$ of length 4 m and area of cross-section 10 cm² is heated from 0° C to 400°C without being allowed to extend. The tension produced in the rod is $x \times 10^5$ N where the value of x is

46. Consider the following statements regarding series grouping of capacitors and select the correct statements.

[NCERT-XII, Page 72]

- I. Charge on each capacitor remains same and equals to the main charge supplied by the battery.
- II. Potential difference and energy distributes in the reverse ratio of capacitance.
- III. Effective capacitance is even less than the least of the individual capacitances.
- IV. Potential on each capacitor remains same and equals to the potential supplied by battery.
- (1) I and II (2) III and IV
- (3) II and IV only (4) I, II and III

~
L
o
v

47. Susceptibility of ferromagnetic substance is

> **[NCERT-XII, Page 147** (3) 0 (1) > 1(2) < 1(4) 1

48. The force is given in terms of time t and displacement x by the equation **INCERT-XI, Page 7, 81** $F = A \cos Bx + C \sin Dt$

The dimensional formula of $\frac{AD}{B}$ is :

- (1) $[M^0 L T^{-1}]$ (2) $[M L^2 T^{-3}]$
- (4) $[M^2 L^2 T^{-3}]$ (3) $[M^1 L^1 T^{-2}]$

49. Two coils have a mutual inductance 0.005H. The current changes in first coil according to equation $I = I_0 \sin \omega t$ where $I_0 = 10A$ and $\omega = 100\pi$ radian/sec. The max. value of e.m.f. in second coil is [NCERT-XII, Page 167] (1) 2π (2) 5π

(3) π (4) 4π

50. When a current I is set up in a wire of radius r, the drift velocity is v_d. If the same current is set up through a wire of radius 2 r, the drift velocity will be [NCERT-XII, Page 86]

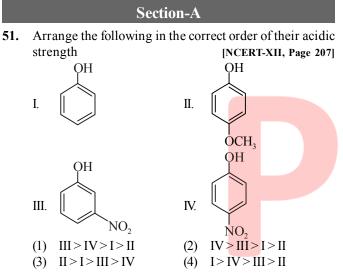
(1)	4 v _d	(2)	$2 v_d$

(3) $v_d/2$ (4) $v_d/4$

PART-II: CHEMISTRY

56.

Η



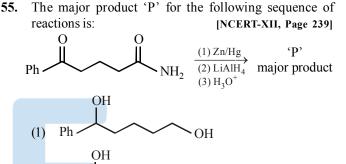
52. Assertion: The stability order of +1 oxidation state of Ga, In and T1 is Ga < In < T1.

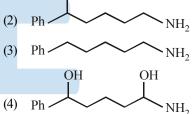
Reason: The inert pair effect stabilizes the lower oxidation state down the group.

- (1) If both Assertion and Reason are correct and the Reason is a correct explanation of the Assertion.
- (2) If both Assertion and Reason are correct but Reason is not a correct explanation of the Assertion.
- If the Assertion is correct but Reason is incorrect. (3)
- If the Assertion is incorrect and Reason is correct. (4)
- 53. The number of molecules in 2.8375 litres of O_2 at STP are respectively [NCERT-XI, Page 18] (1) 7.527×10^{24} (2) 1.505×10^{23}
 - (3) 7.527×10^{23} (4) 7.527×10^{22}
- 54. Statement I: For isothermal irreversible change of an ideal gas, $q = -w = P_{ext} (V_{final} - V_{initial})$ Statement II: For adiabatic change, $\Delta U = W_{adiabatic}$

[NCERT-XI, Page 142]

- Both statement I and II are correct. (1)
- (2)Both statement I and II are incorrect.
- (3)Statement I is correct but statement II is incorrect.
- Statement II is correct but statement I is incorrect. (4)





[NCERT-XII, Page 241]

$$H_{3C} \longrightarrow C = CH_{2} \xrightarrow{(1)O_{3}} X + Y$$

$$H_{3C} \longrightarrow CH_{2} \xrightarrow{(2)Zn/H_{2}O} X + Y$$

$$X + Y \xrightarrow{(1) \text{dil. NaOH}} Z$$

Consider the reactions

The IUPAC name of 'Z' is

- But-1-en-3-one (1)
- 4-Hydroxybutan-2-one (2)
- But-3-en-2-one (3)
- (4) 1-Hydroxybutan-3-one
- 57. Statement I: Liquids A and B form a non-ideal solution with positive deviation. The interactions between A and B are weaker than A-A and B-B interactions Statement II: For an ideal solution. $\Delta_{mix}H = 0:\Delta_{mix}V = 0$

[NCERT-XII, Page 10]

- Both statement I and II are correct. (1)
- Both statement I and II are incorrect. (2)
- (3)Statement I is correct but statement II is incorrect.
- (4) Statement II is correct but statement I is incorrect.

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- **58.** Identify the mixture that shows positive deviations from Raoult's Law [NCERT-XII, Page 10] (1) $CHCl_3 + (CH_3)_2 CO$ (2) $CHCl_3 + C_6H_6$ (3) $(CH_3)_2CO + C_6H_5NH_2$ (4) $(CH_3)_2CO + CS_2$
- **59.** Choose the correct option for free expansion of an ideal gas under adiabatic condition from the following: (1) $q=0, \Delta T \neq 0, w=0$ [NCERT-XI, Page 140]
 - (2) $q = 0, \Delta T < 0, w \neq 0$
 - (3) $q = \neq 0, \Delta T = 0, w = 0$
 - (4) $q=0, \Delta T=0, w=0$
- 60. Match the complexes in list-I with their hybridization in list-II. [NCERT-XII, Page 128]

	List-I (Complex)	Ι	List-II (Hybridization)
(A)	Ni(CO) ₄	(p)	sp^3d^2
(B)	$[Ni(CN)_{4}]^{2-}$	(q)	d^2sp^3
(C)	$[Co(NH_3)_6]^{3+}$	(r)	dsp^2
(D)	$[CoF_{6}]^{3-1}$	(s)	sp^3
(1)	A - (r), B - (s), C - (p)	, D –	(q)
(2)	A - (r), B - (s), C - (q)	, D –	(p)
(3)	A - (s), B - (r), C - (q)	, D –	(p)
(A)	\mathbf{A} (-) \mathbf{D} (-) \mathbf{C} (-)	D	(-)

- (4) A-(s), B-(r), C-(p), D-(q)
- 61. Methods used for purification of organic compounds are based on : [NCERT-XI, Page 278]
 - (1) neither on nature of compound nor on the impurity present.
 - (2) nature of compound only.
 - (3) nature of compound and presence of impurity.
 - (4) presence of impurity only.
- 62. Observe the following structure
 - $\ddot{O} = \ddot{N} \ddot{O}$:
 - (1) (2) (3)

The formal charges on the atoms 1, 2, 3 respectively are

- [NCERT-XI, Page 104] (1) +1,0,-1 (2) 0,0,-1 (2) 0,0,-1
- $(3) \quad -1, 0, +1 \qquad \qquad (4) \quad 0, 0, 0$
- **63.** Statement I : The metallic radius of Na is 1.86 Å and the ionic radius of Na⁺ is lesser than 1.86 Å.

Statement II : Ions are always smaller in size than the
corresponding elements.[NCERT-XI, Page 87]

- (1) Both statement I and II are correct.
- (2) Both statement I and II are incorrect.
- (3) Statement I is correct but statement II is incorrect.
- (4) Statement II is correct but statement I is incorrect.

64. The ratio
$$\frac{K_P}{K_C}$$
 for the reaction
 $CO(g) + \frac{1}{2}O_2(g) \rightleftharpoons CO_2(g)$ is: [NCERT-XI, Pa

$$CO(g) + \frac{1}{2}O_2(g) \rightleftharpoons CO_2(g)$$
 is: [NCERT-XI, Page 178]

(1)
$$(RT)^{1/2}$$
 (2) RT (3) 1 (4) $\frac{1}{\sqrt{RT}}$

5.
$$Cl \xrightarrow{H_2} CHO$$

This reduction reaction is known as :

- Rosenmund reduction [NCERT-XII, Page 232]
- (2) Wolff Kishner reduction
- (3) Stephen reduction
- (4) Etard reduction

6

(1)

66. Statement I: Dimethyl glyoxime forms a six - membered covalent chelate when treated with $NiCl_2$ solution in presence of NH_4OH .

Statement II: Prussian blue precipitate contains iron both in (+2) and (+3) oxidation states.

- (1) Both statement I and II are correct.
- (2) Both statement I and II are incorrect.
- (3) Statement I is correct but statement II is incorrect.
- (4) Statement II is correct but statement I is incorrect.

67. The
$$E^0$$
 of M|M²⁺||Cu²⁺/Cu is 0.3 V, At what concentration
of Cu²⁺(in mol L⁻¹), the E_{cell} value becomes zero?

$$\left(\frac{2.303\text{RT}}{\text{F}} = 0.06\right) \text{ (Conc. of } \text{M}^{2+} = 0.1 \text{ M}\text{)}$$
[NCERT-XII, Page 38]

(1) 10⁻⁹

- (2) 10^{-8}
- $\begin{array}{ccc} (3) & 10^{-11} \\ (4) & 10^{-10} \end{array}$
- 68. The element having the highest first ionization enthalpy is [NCERT-XI, Page 87]
 - (1) C (2) Al
 - (1) C (2) A1 (3) Si (4) N
 - (5) 51 (4) 1
- 69. Assertion: $S_N 2$ reaction of $C_6H_5CH_2Br$ occurs more readily than the $S_N 2$ reaction of CH_3CH_2Br .

Reason: The partially bonded unhybridized p-orbital that develops in the trigonal bipyramidal transition state is stabilized by conjugation with the phenyl ring.

[NCERT-XII, Page 171]

- (1) If both Assertion and Reason are correct and the Reason is a correct explanation of the Assertion.
- (2) If both Assertion and Reason are correct but Reason is not a correct explanation of the Assertion.
- (3) If the Assertion is correct but Reason is incorrect.
- (4) If the Assertion is incorrect and Reason is correct.

70. Isomerisation of gaseous cyclobutene to butadiene is a first order reaction. At T (K), the rate constant of the reaction is $3.3 \times 10^{-4} \, s^{-1}$. What is the time required (in min) to complete 90% of this reaction at the same temperature? (log 2 = 0.3) [NCERT-XII, Page 73]

0	,		[
(1)	116.67	(2)	233.34
(3)	58.34	(4)	350.0

Match Column-I with Column-II

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71.

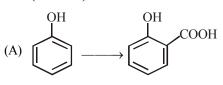
	Column-I		Column-II	
	(Compound)		(Shape)	
	(A) BrF_5	(p)	bent	
	(B) SF_6	(q)	square pyramic	dal
	(C) H ₂ Ŏ	(r)	trigonal bipyra	midal
	(D) PCl_3F_2	(s)	octahedral	
	Choose the correct answe			en below:
	(1) (A) - (p); (B) - (q); (C) - (r); (D) - (s)	
	(2) (A) - (s); (B) - (r); (C)		, u ,	
	(3) (A) - (q); (B) - (s); (C)			
	(4) (A) - (r); (B) - (s); (C)	- (q); (l	D) - (p)	
72.	For a first order reaction was reduced from 0.03 mc What is its rate (in mol L (1) 6.667×10^{-6} (3) 6.667×10^{-4}	$ \begin{array}{c} \text{ol } \mathrm{L}^{-1} \text{ to} \\ \mathrm{L}^{-1} s^{-1})? \\ \text{(2)} \end{array} $	0.02 mol L ⁻¹ ii [NCERT-XI	n 25 min.
		0	1	
		\searrow	1	
73.	Statement I: O ₂ N	II Ind-A	10 ₂	
	IUPAC name of compound		hlor <mark>o-2,4-dinitro</mark>	benzene:
	Statement II:	CH ₃		
	Г С ₂ Н ₅ Сотроилd-В			76
	IUPAC name of Compoun	d B is 4	-ethyl-2-methy NCERT-XI,	
	 Both statement I and Both statement I and Both statement I and Statement I is correct Statement II is correct 	II are i t but sta	correct. ncorrect. atement II is inc	77 correct.
74.	Statements I: Aminobenze	ene and	aniline are same	e organic

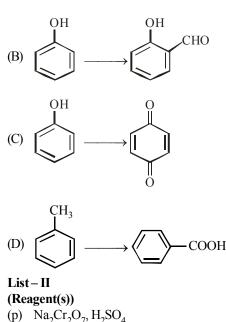
[NCERT-XI, Page 115]

compounds. Statements II: Aminobenzene and aniline are different

organic compounds. [NCERT-XII, Page 261]

- (1) Both statement I and II are correct.
- (2) Both statement I and II are incorrect.
- (3) Statement I is correct but statement II is incorrect.
- (4) Statement II is correct but statement I is incorrect.
- 75. Match List-I with List-II. [NCERT-XII, Page 213] List – I (Reaction)





- (q) (i) $\text{KMnO}_4/\overline{\text{OH}}, \Delta$ (ii) H_3O^+
- (r) (i) NaOH, CHCl₃ (ii) NaOH (iii) HCl
- (s) (i) NaOH(ii) CO₂(iii) HCl

Choose the correct answer from the options given below:

- (1) (A)-(s), (B)-(p), (C)-(r), (D)-(q)
- (2) (A)-(q), (B)-(r), (C)-(p), (D)-(s)
- (3) (A)-(q), (B)-(p), (C)-(r), (D)-(s)
- (4) (A)-(s), (B)-(r), (C)-(p), (D)-(q)

76. Arrange the following in the order of decreasing basicity $RN = CHR^1$ $RC \equiv N$ RNH_2 [NCERT-XII, Page 266]

- $I \qquad II \qquad III$ $(1) I > III > II \qquad (2) III > I > II$
- (1) I > III > II(3) II > III > I(4) II > I > III

7. Match List I with List II: [NCERT-XII, Page 271] List I List II Test Functional group / Class of Compound

- (A) Molisch's Test (p) Peptide
- (B) Biuret Test (q) Carbohydrate
- (C) Carbylamine Test (r) Primary amine
- (D) Schiff's Test (s) Aldehyde
- Choose the **correct** answer from the options given below:
- (1) (A) (p), (B) (q), (C) (r), (D) (s)
- (2) (A) (q), (B) (p), (C) (r), (D) (s)
- (2) (1) (q), (2) (p), (3) (A) (r), (B) (s), (C) (q), (D) (p)
- (4) (A) -(r), (B) -(s), (C) -(p), (D) -(q)

78. Assertion: PH₃ has lower boiling point than NH₃.
 Reason: In liquid state NH₃ molecules are associated through van der Waal's forces, but PH₃ molecules are associated through hydrogen bonding.

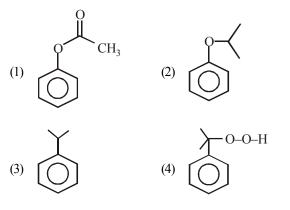
[NCERT-XI, Page 131]

(1) If both Assertion and Reason are correct and the Reason is a correct explanation of the Assertion.

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 - (2) If both Assertion and Reason are correct but Reason is not a correct explanation of the Assertion.
 - (3) If the Assertion is correct but Reason is incorrect.
 - (4) If the Assertion is incorrect and Reason is correct.
- 79. In the cumene to phenol preparation in presence of air, the intermediate is [NCERT-XII, Page 202]



80. Anomalous behavior of oxygen is due to its

[NCERT-XII, Page 187]

- (1) small size and high electronegativity
- (2) small size and low electronegativity
- (3) large size and high electronegativity
- (4) large size and low electronegativity
- 81. In case of isoelectronic species the size of F⁻, Ne and Na⁺ is affected by: [NCERT-XI, Page 87]
 - (1) Principal quantum number (n)
 - (2) None of the factors because their size is the same
 - (3) Electron electron interaction in the outer orbitals
 - (4) Nuclear charge (z)
- 82. Identify correct statements from below:

[NCERT-XII, Page 105]

- A. The chromate ion is square planar.
- B. Dichromates are generally prepared from chromates.
- C. The green manganate ion is diamagnetic.
- D. Dark green coloured K₂MnO₄ disproportionates in a neutral or acidic medium to give permanganate.
- E With increasing oxidation number of transition metal, ionic character of the oxides decreases.

Choose the correct answer from the options given below:

- (1) B, C, D only (2) A, D, E only
- (3) A, B, C only (4) B, D, E only
- 83. But-2-yne is reacted separately with one mole of Hydrogen as shown below: [NCERT-XI, Page 316]

$$\underline{\mathbf{B}} \xleftarrow{\text{Na}}_{\text{liq NH}_3} \mathbf{CH}_3 - \mathbf{C} \equiv \mathbf{C} - \mathbf{CH}_3 \xrightarrow{\text{Pd/C}} \underline{\mathbf{A}}$$
$$+ \mathbf{H}_2$$

- (A) A is more soluble than B.
- (B) The boiling point & melting point of A are higher and lower than B respectively.
- (C) A is more polar than B because dipole moment of A is zero.
- (D) Br_2 adds easily to B than A.

Identify the incorrect statements from the options given below:

- (1) A and B only (2) B, C & D only
- $(3) A, C \& D only \qquad (4) B and C only$
- **84.** If the first ionisation enthalpy of Li, Be and C respectively are 520, 899, 1086 kJ mol⁻¹, the first ionisation enthalpy (in kJ mol⁻¹) of B will be [NCERT-XI, Page 88]
 - (1) 487 (2) 950 (2) 201
 - (3) 801 (4) 1402

85. The **incorrect** statement about Glucose is :

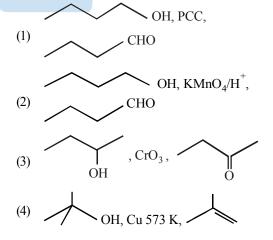
[NCERT-XII, Page 282]

- (1) Glucose is soluble in water because of having aldehyde functional group
- (2) Glucose remains in multiple isomeric form in its aqueous solution
- (3) Glucose is an aldohexose
- (4) Glucose is one of the monomer unit in sucrose

Section-B

- **86.** The sum of oxidation state and co-ordination number of central metal atom is maximum with respect to which of the following complex? [NCERT-XII, Page 119]
 - (1) $K_3[Cr(C_2O_4)_3]$ (2) $[Cr(CO)_6]$ (3) $K_2[PtCl_6]$ (4) $K_4[Fe(CN)_6]$
- 87. An alcohol X ($C_4H_{10}O$) does not give turbidity with conc. HCl and ZnCl₂ at room temperature. X on reaction with reagent Y gives Z. What are X, Y and Z respectively?

[NCERT-XII, Page 208, 210]



88. Observe the following equilibrium at T (K) $H_2(g) + I_2(g) \implies 2HI(g)$ Which one of the following does not disturb the above

which one of the following does not disturb the above equilibrium? [NCERT-XII, Page 185]

- (1) Addition of $H_2(g)$
- (2) Removal of HI(g)
- (3) Addition of $I_2(g)$
- (4) Addition of He(g)

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	v
_	~

89. Identify the factor from the following that **does not** affect electrolytic conductance of a solution.

[NCERT-XII, Page 42]

95.

- (1) The nature of solvent used.
- (2) Concentration of the electrolyte.
- (3) The nature of the electrode used.
- (4) The nature of the electrolyte added.
- Aqueous CuSO₄ solution was electrolysed by passing 2 90. amp of current for 10 min. What is the weight (in g) of copper deposited at cathode?

 $(Cu = 63 \text{ u}; F = 96500 \text{ C mol}^{-1})$

- [NCERT-XII, Page 51] (1) 0.195 (2)0.39 (3) 0.78 (4) 1.56
- **91.** In the reaction sequence Y is

$$\mathrm{CH}_{3}\mathrm{CO}_{2}\mathrm{H} \xrightarrow{(1) \mathrm{NH}_{3}} \mathrm{P} \xrightarrow{\mathrm{Br}_{2}/\mathrm{NaOH}} \mathrm{Y}$$

[NCERT-XII, Page 264]

- (1) a primary amine with same number of carbons as in P
- (2) a primary amine with one carbon less than in P
- (3) a secondary amine with same number of carbons as in P
- (4) a secondary amine with one carbon less than in P
- 92. Observe the following reactions [NCERT-XI, Page 242]

(i)
$$2\text{KClO}_3(s) \xrightarrow{\Delta} 2\text{KCl}(s) + 3\text{O}_2(g)$$

(ii)
$$2H_2O_2(aq) \xrightarrow{\Delta} 2H_2O(l) + O_2(g)$$

(iii)
$$AgNO_3(aq)+KCl(aq) \longrightarrow AgCl(s)+KNO_3(aq)$$

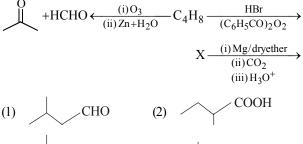
(iv) $2Na(s) + \frac{1}{2}O_2(g) \longrightarrow Na_2O(s)$

The number of redox reactions in this list is

(1)	3	(2)	4	
(3)	2	(4)	1	

- 93. What happens to freezing point of benzene when small quantity of napthalene is added to benzene?
 - (1) Increases [NCERT-XII, Page 18]
 - (2) Remains unchanged
 - (3) First decreases and then increases
 - (4) Decreases
- 94. What is Y in the following reaction sequence?

[NCERT-XII, Page 246]





Match List I with List II	[NCERT-XII, Page 239]
List I	List II
(Name of reaction)	(Reagent used)
(A) Hell–Volhard–	(p) Zn, Hg/HCl

- **Zelinsky** reaction (B) Clemmensen (q) (i) CrO_2Cl_2 , $CS_2(ii)$ H₂O
- (C) Etard reaction
 - (r) (i) Br_2/red phosphorus (ii) H₂O
- (D) Gattermann-Koch (s) CO, HCl, anhyd. AlCl₂ reaction

Choose the correct answer from the options given below:

- (1) A-(r), B-(q), C-(p), D-(s)
- (2) A-(r), B-(p), C-(s), D-(q)
- (3) A-(p), B-(q), C-(r), D-(s)
- (4) A-(r), B-(p), C-(q), D-(s)
- 96. Arrange the following in the increasing order of number of unpaired electrons present in the central metal ion
 - [NCERT-XII, Page 132]

I.	$[Mn Cl_{6}]^{3-}$	II.	$[Fe F_{6}]^{3-}$
III.	$[Mn(CN)_{6}]^{3-}$	IV.	$[Fe(CN)_{6})^{3-}$
	IV <i<iiič<ii< td=""><td>(2)</td><td>I<iii<iĭ<iv< td=""></iii<iĭ<iv<></td></i<iiič<ii<>	(2)	I <iii<iĭ<iv< td=""></iii<iĭ<iv<>
(3)	IV <iii<i<ii< td=""><td>(4)</td><td>I < II < III < IV</td></iii<i<ii<>	(4)	I < II < III < IV

- 97. The fragrance of flowers is due to the presence of some steam volatile organic compounds called essential oils. These are generally insoluble in water at room temperature but are miscible with water vapour in vapour phase. A suitable method for the extraction of these oils from the flowers is -[NCERT-XI, Page 321]
 - (1)crystallisation
 - distillation under reduced pressure (2)
 - (3)distillation

(1) 14.2

- (4) steam distillation
- 98. The density of nitric acid solution is 1.5 g mL^{-1} . Its weight percentage is 68. What is the approximate concentration (in mol L⁻¹) of nitric acid? (N=14 u; O=16 u; H=1 u)
 - [NCERT-XI, Page 23]
 - (2) 11.6 (3) 18.2 (4) 16.2
- 99. Which of the following will make a basic buffer solution. [NCERT-XI, Page 203]
 - 100 mL of 0.1 M CH₃COOH + 100 mL of 0.1 M NaOH (1)
 - 100 mL of 0.1 M HCl + 100 mL of 0.1 M NaOH (2)
 - 50 mL of 0.1 M KOH+ 25 mL of 0.1 M CH₂COOH (3)
 - (4) $100 \text{ mL of } 0.1 \text{ M HC1} + 200 \text{ mL of } 0.1 \text{ M NH}_{4} \text{OH}$

100. Which among the following halide/s will not show S_{N1} reaction: [NCERT-XI, Page 174]

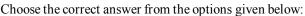
(A) $H_2C = CH - CH_2CI$ (B) $CH_3 - CH = CH - Cl$ CH₂-Cl (D) (C)

CH2-Cl

Choose the most appropriate answer from the options given below:

- (1) (A), (B) and (D) only (2) (A) and (B) only
- (3) (B) and (C) only (4) (B) only

FT - 01 11 **PART-III: BOTANY 109.** Given below are two statements: Section-A Statement-I: The flower of guava is hypogynous. 101. The eukaryotes include-[NCERT-XI, Page 91] Statement-II: In hypogynous flower, the ovary occupies (2) Protists (1) Fungi the lower position. [NCERT-XI, Page 63] (3) Plants and animals (4) All In the light of the above statements, choose the correct answer from the options given below: 102. Read the given statements: Both Statement I and Statement II are true. (1)A. Outer most covering of whole plant body Both Statement I and Statement II are false. (2)Comprises stomata, trichomes and hairs B. Statement I is true but Statement II is false. (3)C. Cells - compactly arranged (4)Statement I is false but Statement II is true. D. Often single layered E Protective tissue [NCERT-XI, Page 71-72] 110. Match List - I with List - II [NCERT-XI, Page 101] The characters are shown by which of the following tissues-List - I List - II (1) Epidermis (2) Collenchyma A. Metacentric L Centromere situated Sclerenchyma (4) Vascular tissue (3)chromosome close to the end forming one extremely short 103. Arrangement of leaves on the stem or branch is called and one very long arms [NCERT-XI, Page 61] II. Centromere at the B Acrocentric Phyllotaxy (2) Ptyxis (1)chromosome terminal end Vernation (4) Venation (3)C. Submetacentric III. Centromere in the middle **104.** Solanum tuberosum is the scientific name of : chromosome forming two equal [NCERT-XI, Page 6] arms of chromosomes (2) Tomato (1) Potato D. Telocentric IV. Centromere slightly (3) Brinjal (4) Lemon chromosome away from the middle forming one shorter **105.** Which of the following is not an exception of cell theory? arm and one longer arm [NCERT-XI, Page 87] Choose the correct answer from the options given below: (1) Bacteria (2) Viruses (1) A-III, B-I, C-IV, D-II(3) Prions (4) Viroids (2) A-I, B-III, C-II, D-IV**106.** Identify the kind of phyllotaxy shown in the given figures (3) A-II, B-III, C-IV, D-IA, B, and C. [NCERT-XI, Page 61] (4) A-I, B-II, C-III, D-IV111. The group of organisms oxidising various inorganic substance such as nitrates, nitrites and ammonia using the released energy for their ATP production are -Chemoheterotrophic bacteria [NCERT-XI, Page 13] (1)Chemosynthetic autotrophic bacteria (2)Photoautotrophic bacteria (3)(4) Saprophytic bacteria **112.** Which of the following is not a part of epidermal tissue В system? [NCERT-XI, Page 71] (1) A - Alternate, B - Opposite, C - Whorled (1) Trichomes (2) Companion cells (2) A - Whorled, B - Opposite, C - Alternate (3) Guard cells (4) Subsidiary cells (3) A - Alternate, B - Whorled, C - Opposite (4) A - Whorled, B - Alternate, C - Opposite 113. The members of class chlorophyceae are commonly called-[NCERT-XI, Page 26] **107.** Which ones forms bloom in polluted water mostly? (1) Red algae (2) Blue green algae [NCERT-XI, Page 13] (3) Green algae (4) Brown algae (1) Cyanobacteria (2) Green algae (3) Red algae (4) Aquatic and terrestrial 114. Match the column-I and column-II. [NCERT-XI, Page 59-60] animals Column-I Column-II Stem Monocot leaf A. I. 108. Which of the following is not multilayer in dicot root? II. Dicot leaf В Leaf [NCERT-XI, Page 74] C. Reticulate venation III. Axillary bud (1) Epidermis (2) Cortex D. IV. Nodes and internodes Parallel venation (4) Pith (3) Pericycle



- (1) A-IV, B-III, C-II, D-I
- (2) A-IV, B-I, C-II, D-III
- (3) A-II, B-IV, C-I, D-III
- (4) A-III, B-IV, C-II, D-I
- 115. Which of the following statement (s) are correct?
 - A. Respiration refers to the release of considerable amount of energy by the hydrolysis of the 3-C bonds of complex compounds through oxidation within the cells.
 - Plants do not need any specialised organs for В gaseous exchange
 - C. In stems, the 'living' cells are organised in thin layers inside and beneath the bark.
 - D. During the process of respiration, oxygen, carbon dioxide, water and energy are released as products.

[NCERT-XI, Page 153-154]

Choose the correct answer from the options given below:

- (1) A, B and C only
- (2) A and D only
- (3) A, C and D only
- (4) B, C and D only
- 116. Given below are two statements:

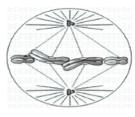
Statement I: DNA replication and centriole replication occur during S-phase in animal cell.

Statement II: Significance of mitosis is in producing cells genetically dissimilar to parent cell.

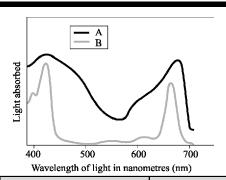
[NCERT-XI, Page 121-125]

In the light of the above statements, choose the correct answer from the options given below:

- (1) Both Statement I and Statement II are true.
- Both Statement I and Statement II are false. (2)
- (3) Statement I is true but Statement II is false.
- (4) Statement I is false but Statement II is true.
- 117. A stage of mitosis is shown in the given diagram. Identify stage with its characteristics? [NCERT-XI, Page 123]



- (1) Late prophase Chromosomes move to spindle equator.
- (2) Metaphase Spindle fibres attached to kinetochores, centromeres split and chromatids separate
- (3) Metaphase chromosomes moved to spindle equator chromosomes made up of two sister chromatids
- (4) Anaphase centromeres split and chromatids separate and start moving away
- **118.** Study the given graph which shows the action spectrum of A superimposed on B spectrum of chlorophyll a. Identify A & B in the graph. [NCERT-XI, Page 137]



	Α	В
(1)	Rate of respiration	Action spectrum
(2)	Rate of respiration	Absorption
(3)	Rate of photosynthesis	Action spectrum
(4)	Rate of photosynthesis	Absorption

- 119. Which is the longest phase in the cell cycle of human liver cells? [NCERT-XI, Page 121]
 - (2) Prophase (1) Anaphase
 - Interphase (4) Telophase (3)
- 120. Match the Column I with Column II-

A.

B.

C.

D.

	Julii I With	Coru	
			[NCERT-XI, Page 65-66]
Colum	I		Column II
Ovary		I.	Fruit
Ovule		II.	Guava, orange, mango
Wall of	ovary	III.	Pericarp
Fleshy	fruit	IV.	Seed

Choose the correct answer from the options given below:

- (1) A-I, B-IV, C-III, D-II
- (2) A-I, B-II, C-III, D-IV
- (3) A-I, B-III, C-II, D-IV
- (4) A-III, B-IV, C-I, D-II
- 121. A normal girl, whose mother is haemophilic marries a male with no ancestral history of haemophilia. What will be the possible phenotypes of the offsprings?

[NCERT-XII, Page 72]

- A. Haemophilic son and haemophilic daughter
- B. Haemophilic son and carrier daughter
- C. Normal daughter and normal son.
- D. Normal son and haemophilic daughter.

Choose the correct answer from the options given below:

- (1) B and D only (2) A and B only
- (3) B and C only (4) A and D only
- 122. According to the fluid mosaic model of the cell membrane, the proteins are located-[NCERT-XI, Page 94]
 - (1) In a continuous layer over the outer surface of the membrane only
 - (2) In a continuous layer over the inner surface only
 - (3) In discontinuous arrangement, both on the surface (as peripheral proteins) and in the interior of the membrane (as integral proteins)
 - (4) In the middle of the membrane, between the lipid layers only

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- 123. Oxygen which is liberated during photosynthesis comes from- [NCERT-XI, Page 139]
 - (1) Carbon dioxide (2) Water
 - (3) Chlorophyll (4) Phosphoglyceric acid
- **124**. Go through the following statements.
 - [NCERT-XI, Page 126-127]
 - A. Meiosis II is similar to mitosis
 - B. Terminalization of chiasmata occurs in meiosis II.
 - C. Meiosis II performs separation of homologous chromosomes.
 - D. Interphase is intermitosis
 - E After mitosis, the number of chromosomes in daughter cells shall one fourth of parent cell.
 - Choose the correct answer from the options given below:

 - (3) A, C and D (4) A and D
- **125.** Read the following four statements, A, B, C and D and select the right option having both correct statements
 - A. Z scheme of light reaction takes place in presence of PS-I only
 - B. Only PS-I is functional in cyclic photophosphorylation
 - C. Cyclic photophosphorylation results into synthesis of ATP and NADPH₂
 - D. Stroma lamellae lack PS-II as well as NADP reductase. [NCERT-XI, Page 139]
 - Choose the correct answer from the options given below:
 - (1) A and B (2) B and C
 - $(3) C and B \qquad (4) B and D$
- **126.** Which of the following induces fruit set in pineapple?
 - (1) Auxin (2) Gibberellin
 - (3) Kinetin (4) Ethylene
- 127. In the first step of monohybrid cross experiment, Mendel selected pea plants which were [NCERT-XII, Page 54]
 - (1) Pure tall as male and pure dwarf as female
 - (2) Pure tall as female and pure dwarf as male
 - (3) Heterozygous tall as male and pure dwarf as female
 - (4) Heterozygous tall as female and pure dwarf as male
- **128.** Match the columns: [NCERT-XII, Page 67, 69, 74] Column A Column B A. Non parental gene I. Crossing over combination Non sister chromatids II. X and Y R C. Sex chromosomes III. Sex-linked disease D. Haemophilia IV. Recombination Choose the correct answer from the options given below: (1) A-II, B-IV, C-I, D-III
 - (2) A-IV, B-I, C-II, D-III
 - (3) A-II, B-IV, C-III, D-I
 - (4) A-II, B-I, C-IV, D-III
- 129. A gene may have many alleles, but each individual has only two alleles because [NCERT-XII, Page 56]
 - (1) having more than two alleles is lethal.

[NCERT-XII, Page 61]

- (2) having more than two alleles unbalances the chromosomes.
- (3) a person has two parents who each contribute one allele.
- (4) a backup set of alleles is necessary in case something goes wrong.
- 130. Root hair development is promoted by [NCERT-XI, Page 175]
 - (1) Abscisic acid
 - (2) Auxin
 - (3) Gibberellin
 - (4) Cytokinin
- **131.** The PGR which causes apical hook formation in dicot seedling also helps in

 [NCERT-XI, Page 177]
 - A. Overcoming the apical dominance
 - B. Widely used as herbicides
 - C. Delays senescence
 - D. Induces flowering in mango
 - E Also known as antigibberellin
 - Choose the correct answer from the options given below:
 - (1) A and C (2) B and D
 - (3) D only (4) D and E
- **132.** In the AB blood group the two genes are
 - (1) codominant
 - (2) corecessive
 - (3) incompletely dominant
 - (4) dominant-recessive
- 133. Linkage was discovered by [NCERT-XII, Page 67]
 - (1) Mendel (2) Morgan
 - (3) Crick (4) Muller
- 134. During the process of respiration, which of the following
is not released?[NCERT-XI, Page-154 155]
 - (1) Carbon dioxide
 - (2) Water
 - (3) Oxygen
 - (4) Energy

(1) Brazil

- **135.** Given below are two statements:
 - Statement I: A diploid female plant and a tetraploid male plant are crossed. The ploidy of endosperm shall be 3N. Statement II: Milky water of green coconut is liquid endosperm. [NCERT-XII, Page 18-19] In the light of the above statements, choose the correct answer from the options given below:
 - (1) Both Statement I and Statement II are true.
 - (2) Both Statement I and Statement II are false.
 - (3) Statement I is true but Statement II is false.
 - (4) Statement I is false but Statement II is true.

Section-B

- 136. Which of the following countries has the highest biodiversity? [NCERT-XII, Page 219]
 - (2) South Africa
 - (3) Russia (4) India

[NCERT-XII, Page 93, 94]

1	4

- 137. In an area, there are 200 *Parthenium* and a single huge banyan tree. Which of the following conclusions is correct? [NCERT-XII, Page 192]
 - (1) Population density of banyan is low relative to that of *Parthenium*.
 - (2) Population cover area of banyan is high relative to *Parthenium*.
 - (3) In the above case% cover or biomass is a more meaningful measure of the population size.
 - (4) All the above
- **138.** Experiments by Avery, MacLeod, and McCarty supported DNA as the genetic material by showing that

[NCERT-XII, Page 85]

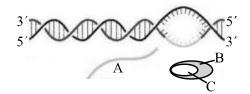
- (1) both protein and DNA samples provided the transforming factor
- (2) DNA was not complex enough to be the genetic material
- (3) only samples with DNA provided transforming activity
- (4) even though DNA was molecularly simple, it provided adequate variation to act as the genetic material
- **139.** Information transfer from RNA to DNA is called
 - (1) Replication (
- [NCERT-XII, Page 83]
- (3) Translation (4)
- (2) Reverse transcription
 - slation (4) Trans<mark>criptio</mark>n
- **140.** Given below are two statements:

Statement I: India is divided into ten biogeographical regions.

Statement II: India is with only 2.4% of the world's land area and possesses 8.1% species diversity of the world.

[N<mark>CERT</mark>-XII, Page 219]

- In the light of the above statements, choose the correct answer from the options given below:
- (1) Both Statement I and Statement II are true.
- (2) Both Statement I and Statement II are false.
- (3) Statement I is true but Statement II is false.
- (4) Statement I is false but Statement II is true.
- 141. The given figure represent one of the step in the process of transcription in bacteria. Identify the step and label A, B & C marked in the figure. [NCERT-XII, Page 93]



- (1) Initiation; A-DNA, B-RNA, C-Promoter
- (2) Termination; A RNA, B RNA polymerase, C – Rho factor
- (3) Elongation; A RNA, B RNA polymerase, C – Sigma factor
- (4) Elongation; A DNA, B DNA polymerase, C-RNA

- Column-I Column-II A. Exon L RNA of influenza virus II. Functional DNA B. Intron C. Genetic RNA III. RNA of eukarvotes Non-genetic RNA IV. Junk DNA D. Choose the correct answer from the options given below: (1) A-II, B-III, C-I, D-IV(2) A-II, B-IV, C-I, D-III(3) A-IV, B-III, C-I, D-II(4) A-II, B-I, C-III, D-IV**143.** Match the columns : [NCERT-XII, Page 223] Column-I Column-II
 - A. Parthenium I. Lake victoria
 - B. *Nile Perch* II. Water hyacinth
 - C. *Eichhornia* III. African catfish
 - D. Clarias gariepinus IV. Carrot grass

Choose the correct answer from the options given below:

(1) A-I, B-III, C-IV, D-II

142. Match the columns :

- (2) A-II, B-IV, C-III, D-I
- $(3) \quad A-IV, B-I, C-II, D-III$
- (4) A-IV, B-I, C-III, D-II
- 144. Which statement best describes what ultimately happens to the chemical energy that is converted into new biomass in the process of energy transfer between trophic levels in an ecosystem? [NCERT-XII, Page 209]
 - (1) It is used by organisms to maintain their life processes through the reactions of cellular respiration
 - (2) It is undigested and winds up in the faeces and is not passed on to higher trophic levels
 - (3) It is eliminated as faeces consumed in ingestion, respiration or is dissipated into space in accordance with the second law of thermodynamics
 - (4) None of food chain

145. Which of the following coined the term linkage?

[NCERT-XII, Page 67]

- (1) Morgan
- (2) Mendel
- (3) Darwin
- (4) None of them

146. Match the columns : Column-I

AUG

UCU

Α

B.

C.

D.

[NCERT-XII, Page 96] Column-II

- I. Phenyl alanine
- II. Methionine
- UGG III. Serine
- UUU IV. Tryptophan

Choose the correct answer from the options given below:

- (1) A-I, B-II, C-IV, D-III
- (2) A-IV, B-I, C-III, D-II
- $(3) \quad A-II, B-IV, C-I, D-III$
- (4) A-II, B-III, C-IV, D-I

147. Match the Column-I with the Column-II.

Column-I

A. The unequivocal

proof that DNA is

Transformation

the genetic material

[NCERT-XII, Page 84, 85, 89]

Column-II

I.

Frederick Griffith

II. Post-transcription

protein

The mycorrhizal association of fungus have

importance in the absorption of essential nutrients.

The mycorrhizal association is a mutualistic symbiosis

Plants provide the fungi with energy-yielding

D. Fungal partner is associated with the only roots of

Choose the correct answer from the options given below:

the higher plants (like angiosperms)

modification of

FT - 01

R

В

C.

E.

species.

proteins.

(3) Only A and C

(4) Only B and E

(1) OnlvE

(2) Only D

Section-A 151. Read the given statements. [NCERT-XI, Page 42] A. Extracellular and intracellular digestion

- Exclusively marine, radially symmetrical, diploblastic, B tissue level of organization
- C. Bisexual, external fertilization and indirect development
- D. No asexual reproduction
- Choose the correct answer from the options given below:
- (2) Porifera (1)Cnidaria
- (4) Rotifers (3) Ctenophora

152. Which of the following is not true about frogs?

- [NCERT-XI, Pages 81-82] A. Alimentary canal of frogs is short
- Undigested wastes passes out through cloaca B.
- C. Respiration by lungs i.e., pulmonary respiration

	C. D.	DNA replicates semiconservatively Ribozyme action		Alfred Hershey and Martha Chase Matthew Meselson and	(3)	$\begin{array}{l} A-I, B-II, C-IV, I\\ A-I, B-III, C-IV,\\ A-III, B-II, C-I, I\end{array}$	D – I	Ι
		ose the correct answer A - II, B - IV, C - I, D		Franklin Stahl. m the options given below:	150. Mat	tch the columns : Column-I		[NCERT-X Column-]
	(2) (3)	A-III, B-I, C-IV, I A-III, B-I, C-II, I A-IV, B-III, C-I, I)-])-[II V	А.	Exponential growth equation	I.	$\frac{dN}{dt} = rN$
148.	~ /	ich of the following st e?	aten	nents about mycorrhizae is [NCERT-XII, Page 201-202] fits on both the interacting	B.	Integral form of exponential growth equation	II.	$N_t = N_0 e^{rt}$

149. Match the columns :

Α

B.

C.

D.

Column-I

Tropical rain forest

Narrowly utilitarian

(1) A-I, B-III, C-II, D-IV

Steller's sea cow

Water hyacinth

- C. Logistic growth equation
 - Integral form of D logistic growth equation

$$IV. \quad N_t = \frac{K}{1 + \left(\frac{K - N_0}{N_0}\right)}e^{-rt}$$

III. $\frac{dN}{k} = rN\left(\frac{K-N}{K}\right)$

Choose the correct answer from the options given below:

- (1) A-II, B-I, C-IV, D-III
- (2) A-I, B-II, C-III, D-IV
- (3) A-I, B-IV, C-III, D-II
- (4) A-I, B-IV, C-II, D-III
- D. Vascular system open type
 - Heart has three chambers E
 - Choose the correct answer from the options given below:
 - (2) Only E

 - (4) Only C and E (3) Only A and B
- 153. The 20 different common amino acids have different-
 - [NCERT-XI, Page 143]
 - **R**-groups (2) Acid groups (1)(4) 1° structure (3) Peptide bonds
- 154. Match the columns: [NCERT-XI, Pages 156, 158, 146] Column-I Column-II
 - Mg^{2+} Cytochrome c oxidase I. A
 - RuBisCo B C. Hexokinase

Pyruvate

D

III. Two copper centre

II. Bundle sheath cell

- IV. Glycolysis
- dehydrogenase

PART-IV: ZOOLOGY

15

[NCERT-XII, Page 222-223]

[NCERT-XII, Page 194-195]

Column-II

III. Over exploitation

6% cover

IV. Alien species

invasion

Column-II

II. Firewood

L

Choose the correct answer from the options given below:

- (1) OnlvD

16	
10	

Choose the correct answer from the options given below:

- (1) A-III, B-II, C-IV, D-I
- (2) A-II, B-I, C-IV, D-III
- $(3) \quad A-III, B-I, C-II, D-IV$
- (4) A-I, B-IV, C-III, D-II
- **155.** Given below are two statements: one is labelled as Assertion A and the other is labelled as Reason R: **Assertion A:** Oxygen can bind with haemoglobin in a

irreversible manner. **Reason R:** Binding of oxygen with haumoglobin is primarily related to partial pressure of O_2 . [NCERT-XI, Page-189] In the light of the above statements, choose the correct answer from the options given below:

- (1) Both A and R are true and R is the correct explanation of A.
- (2) Both A and R are true but R is NOT the correct explanation of A.
- (3) A is true but R is false.
- (4) A is false but R is true.

	(.)					
156.	Mat	ch the columns : Column-I		[NCERT-XI, Page 203] Column-II		
	٨		т			
	A.	Angina pectoris	I.	Heart does not pump		
				blood effectively enough		
	ъ	TT 4 44 1	п	to meet the needs of body		
	B.	Heart attack	II.	Acute chest pain when no		
				enough oxygen reaches		
	C	II (C.1		heart		
	C.	Heart failure	III.	Damage of heart muscles		
				suddenly by inadequate		
	-	a		blood supply		
	D.	Cardiac arrest		Heart stops beating		
				n the o <mark>ptions</mark> given below:		
	· · ·	A–I, B–IV, C–III,				
		A–II, B–III, C–I, I				
	· /	A-II, B-IV, C-I, D			16	60 .
	(4)	A–I, B–III, C–II, I)–ľ	V		
157.	Mat	ch the columns :		[NCERT-XI, Page 206]		
		Column-I		Column-II		
	A.	Protonephridia/	I.	Cockroach		
		Flame cells				
	B.	Nephridia	Π.	Prawns		
	C.		III.	Planaria		
	D.	10				
	2.	Green glands				
	Cho	-	froi	n the options given below:		

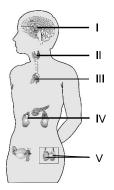
Choose the correct answer from the options given below:

- (1) A-II, B-I, C-IV, D-III
- (2) A-I, B-IV, C-II, D-III
- $(3) \quad A-III, B-I, C-II, D-IV$
- (4) A-III, B-IV, C-I, D-II

158. Match the columns and find out the correct combination:

			[NCERT-XI, Page 224]
	Column-I		Column-II
А.	Stapes	I.	Ear ossicle
B.	Occipital	II.	Facial bone
C.	Hyoid	III.	Cranial bone
D.	Zygomatic	IV.	U-shaped

- Choose the correct answer from the options given below:
- (1) A-I, B-III, C-IV, D-II
- (2) A-II, B-III, C-I, D-IV
- (3) A III, B II, C I, D IV
- (4) A-II, B-I, C-III, D-IV
- 159. The given diagram represents the location of human endocrine glands I, II, III, IV and V. [NCERT-XI, Page 240]



Which of the following gland is correctly matched with their secretions?

Hormones	Their secretions
Ι	Melatonin
II	Thymosin
III	Epinephrine
IV	Aldosterone
V	Testosterone

I, II and III only
 I, IV and V only
 II, IV and V only
 II, III and V only

160. Given below are two statements: one is labelled as Assertion A and the other is labelled as Reason R:

Assertion A: The changes in the ovarian cycle as well as uterine cycle are induced by changes in the level of pituitary and ovarian hormones.

Reason R: The secretion of LH and FSH increases gradually during the follicular phase and estrogen also secreted by developing follicle. [NCERT-XII, Pages 34-35] In the light of the above statements, choose the correct answer from the options given below:

- (1) Both A and R are true and R is the correct explanation of A.
- (2) Both A and R are true but R is NOT the correct explanation of A.
- (3) A is true but R is false.
- (4) A is false but R is true.

161. Given below are two statements:

Statements I: Oral contraceptive pills prevent ovulation. Statements II: It contains progestogen-estrogen combinations. [NCERT-XII, Page 45] In the light of the above statements, choose the correct answer from the options given below:

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(2) E (3) S	Both Statement I and State Both Statement I and State Statement I is true but Sta Statement I is false but Sta	ement II are false. tement II is false.	167. Systemic circulation of oxygenated blood starts from [NCERT-XI, Page 201](1) Right atrium (3) Left ventricle(2) Left atrium (4) Right ventricle
(C olumn-I .amarck I.	Column-II Theory of inheritance of acquired characters or the theory use and disuse of organ	 168. Which is the correct pathway for passage of urine in humans? [NCERT-XI, Page 207] (1) Collecting tubule → ureter → bladder → urethra (2) Renal vein → renal ureter → bladder → urethra (3) Pelvis → Medulla → bladder → urethra (4) Cortex → Medulla → bladder → ureter
C. E D. F Choos (1) A (2) A (3) A (4) A	Ernst HaeckelII.DarwinIII.Hugo de VriesIV.te the correct answer from the cor	Embryological support for evolution Theory of natural selection Mutation theory the options given below:	 (4) Contex → Incluina → onader → arcter 169. Match the columns [NCERT-XI, Page 236] Column-I Column-II A. Association areas I. Limbic lobe B. Hypothalamus II. Body temperature C. Hippocampus III. Intersensory associations D. Medulla IV. Cardiovascular reflexes Choose the correct answer from the options given below: (1) A-I, B-III, C-IV, D-II (2) A-II, B-III, C-I, D-IV (3) A-III, B-II, C-I, D-IV
earthw (1) P (3) T 164. Which A. F B. T C. T D. T Choos (1) C	vorms is – Parasitism (2) Friploblastic (4) In of the following stateme Frog belongs to class amp They do not have constar They are warm blooded. They show camouflage se the correct answer from Dnly A and D (2)	[NCERT-XI, Pages 38-39] Acoelomate nature Pseudocoelomate ents about frogs are false- bhibia [NCERT-XI, Page-80]	 (d) A – II, B – I, C – III, D – IV (4) A – II, B – I, C – III, D – IV 170. Which of the following hormones is mismatched with stated functions? [NCERT-XI, Page 242] (1) ACTH - Stimulates the adrenal cortex (2) Oxytocin - Stimulates water reabsorption by the kidneys (3) Prolactin (PRL) - regulates the growth of mammary glands and milk production in them (4) TSH - Promotes the synthesis and secretion of thyroid hormones from thyroid gland 171. Which of the following hormone levels will cause release of ovum (ovulation) from the Graafian follicle?
A. C c B. C C. N D. C Choos	oenzymes and metal ions Cofactors are tightly bour VAD and NADP contain t Catalytic activity is regain se the correct answer from	ified: prosthetic groups, s. ad to the apoenzyme.	 [NCERT-XII, Page 32] (1) High concentration of progesterone (2) Low concentration of LH (3) Low concentration of FSH (4) High concentration of oestrogen 172. The given diagram shows the uterine tubes of four women (P, Q, R and S). [NCERT-XII, Page 45]
(3) C 166. Match A. F B. h V C. T	a the columns w.r.t. respin Column-I Residual volume I. nspiratory reserve II. rolume	A and C catory capacities of lungs: NCERT-XI, Pages 186-187] Column-II 2500 mL – 3000 mL 3000 mL – 3500 mL 500 mL 1100 mL – 1200 mL	P partial blockage Q complete blockage R complete blockage S oviduct cut and tied

Choose the correct answer from the options given below:

- (1) A-I, B-IV, C-II, D-III
- (2) A-III, B-I, C-II, D-IV
- (3) A-IV, B-III, C-I, D-II
- (4) A-IV, B-I, C-III, D-II

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T	σ

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	(1)	P and Q	(2)	tion impossible at present? Q and R	179.		tch the columns : Column-I	Ţ	[NCERT-XII, Page 122, 123] Column-II
173.	()	R and S ch the columns :	(4)	S and P [NCERT-XII, Page 122]		А. В.	500 mya 350 mya	I. II.	Sea weeds existed Invertebrates were formed and active
	A.	Column-I Cretaceous	I.	Column-II Amphibians dominant, origin of reptiles, extensive forests of		C. D.	320 mya 200 mya	IV.	Reptiles of different shapes and sizes dominated Jawless fishes evolved
	B.	Jurassic	Π.	vascular plants Age of fishes, origin of amphibians		(1)	A–II, B–I, C–IV	, D – I	
	C.	Carboniferous		Angiosperms appear, dinosaurs disappear			A – III, B – II, C – I A – IV, B – II, C – I A – II, B – IV, C – I	, D – I	II
	D.	Devonian		Dinosaurs and gymnosperms dominate	180.	Wh	ich mosquito acts as Anopheles	s vecto	
	(1) (2) (3)	$\begin{array}{l} A-III, B-IV, C-I, \\ A-III, B-II, C-IV, \\ A-I, B-II, C-III, \end{array}$, D – V, D – D – I	- I V	181.	(3)	-		All [NCERT-XI, Page 80] Column-II
174.	. /	A-III, B-IV, C-II er exits from sponges	·	ugh the-		A. B. C.	Protective coloration Summer sleep Winter sleep	II.	Tympanum Mimicry Aestivation
	(1) (3)	Osculum Spicules		[NCERT-XI, Page 40] Ostia Choancytes		D. Cho	Ear Dose the correct answ	IV. ver fro	Hibernation m the options given below:
175.	(1)	<i>a tigrina</i> is zoologica Frog Lizard	(2)	ne of INCERT-XI, Page 80 Leopard Earthworm		 (1) (2) (3) (4) 	A – I, B – III, C – II A – II, B – III, C – I A – II, B – III, C – I A – II, B – IV, C – I	V, D – , D – I	I V
176.		ch the columns: Column-I Transferase Lyase	I. II.	[NCERT-XI, Page 117] Column-II $S-G+S' \rightarrow S+S'-G$ Removal of groups	182.	A. B.	Column-I Insulin Lecithin	I. II.	
	(1)	Ligase Hydrolase ose the correct answe A-I, B-II, C-III, T A-II, B-I, C-IV, T	er fro D – I			(1)	Fructose Anticoagulant oose the correct answ A-II, B-III, C-I A-IV, B-III, C-	IV. ver fro , D−I	
	(3)	A–I, B–III, C–II, A–IV, B–III, C–I,	D-Γ	V		(3)	A IV, B III, C A-IV, B-II, C-I A-II, B-III, C-I	II, D –	Ι
177.				erised by the presence of [NCERT-XI, Page 236]	183.	stat	ements		s and identify the incorrect [NCERT-XII, Pages 27, 28]
	(1) (3)	Optic lobes Cerebellum	· · ·	Corpus callosum Cerebrum		A. B.	Theca cells is prese Temperature in scrusshould be 2°C below	tum ne	ecessary for sperm formation
178.	Mat A. B. C.	ch the columns: Column I Semen Birth canal Penis	I. II. III.	Column II Clitoris Testicular lobules Vagina		C. D. E.	Common duct form and duct of semina Prostate, Bulbourd accessory glands of	ned by il vesic ethral of male	y the union of vas deferens cle is ejaculatory duct. and seminal vesicles are e reproductive system. nnected with the abdominal
	(1) (2) (3)	Seminiferous tubule ose the correct answer A-II, B-I, C-IV, I A-IV, B-II, C-III A-IV, B-III, C-II A-IV, B-III, C-I, A-IV, B-III, C-I, I	er from D – II [, D – [, D –	n the options given below: I I I		Cho (1) (3)	cavity by ligament pose the correct answ A, B, C and E only C, D and E only	ver fro (2)	m the options given below: A and E only A, B, D and E only.

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 184. Given below are two statements: Statements I: Continental drift caused survival of pouched mammals of Australia. Statements II: It was due to lack of competition from any other mammal. [NCERT-XII, Page 117] In the light of the above statements, choose the correct answer from the options given below: Both Statement I and Statement II are true. Both Statement I and Statement II are false. Statement I is true but Statement II is false. 			189.	 Given below are two statements: Statements I: In PCR, the multiple copies of gene of interest is synthesized in vivo. Statements II: Primers used in PCR are small synthetic oligonucleotides. [NCERT-XII, Pages 172-173] In the light of the above statements, choose the correct answer from the options given below: Both Statement I and Statement II are true. Both Statement I and Statement II are false. Statement I is true but Statement II is false. 			
 D. Produ Choose the (1) A-III (2) A-IV (3) A-IV 	n I I. II. tic fluid III. ction of ova IV.	Oogenesis Located in scrotum Labia majora m the options given below: I I I	190.	A. B. C. D. Cho (1) (2) (3)	Microinjection Cloning vector	II. III. IV. from D - IV D - IV D - IV	V V
	Section-		191.		d the following staten		
Statement inflammato Statements In the ligh answer from (1) Both S (2) Both S (3) Statem (4) Statem 187. Match the c	w are two statemen s I: Histamine is ry reactions. II: Histamine is va to of the above state n the options given tatement I and Stat tatement I and Stat nent I is true but Stat nent I is false but Stat olumns w.r.t. HIV-e	ts: involved in allergic and sodilator. [NCERT-XI, Page 194] ements, choose the correct a below: tement II are true. tement II are false. atement II is false. atement II is true. nzymes and their functions: NCERT-XII, Pages 134, 135]		 A. B. C. D. E. How (1) (3) 	Malignant tumors are cells called neoplasti Malignant tumor s competing for vital n Cells of malignant tu The cells of malign property of contact in The cells sloughed fr	e the c cel starv utrie mors ant t nhibi om b wher tatem (2) (4)	mass of non- proliferating ls ves the normal cells by ints s show metastasis cumor severely looses the tion benign tumor reach distant e they get lodged to start a nents are correct? Four Three
B. Physic C. Cellula D. Cytoki Choose the (1) A - I, (2) A - II, (3) A - II,	al barrier I. logical barrier II. rr barrier III. ne barrier IV.	V V		Stat direc Stat of de In th answ (1) (2) (3) (4)	ements I: Darwinia: ctionless. ements II: Evolution i etermination. he light of the above wer from the options g Both Statement I and Both Statement I and Statement I is true bu Statement I is false bu	n va is a d state given Stat Stat Stat t Sta ut Sta ut Sta	riations are random and irected process in the sense [NCERT-XII, Page-119] ments, choose the correct below: ement II are true. tement II are false. itement II is false. atement II is true.
Australian (1) Adapt (2) Adapt	continent. This rep ive radiation like in ive radiation unlike	[NCERT-XII, Page 117]	193.		A technology from the	follo nd ex oter.	

- (3) Convergent evolution like in Darwin's finches
- (4) Convergent evolution unlike in Darwin's finches
- D. One or more identifiable marker genes. E One or more unique restriction sites.

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 Choose the correct answe (1) A, C, D and E only (2) A, C and E only (3) B, C, D and E only (4) B, C and E only 	r from the options given below:	1	Match the columns: Column-I A. Stanley Cohen and Herbert Boyer B. Ethidium bromide	I.	CERT-XII, Pages 164, 168, 169] Column-II Selectable marker Isolated the antibiotic
 194. Read the following statem A. Very low concentrati when the symptoms of can be detected by an by PCR B. PCR is now routinely AIDs patients 	on of bacteria or virus (at a time of the disease are not yet visible) aplification of their nucleic acid used to detect HIV in suspected o detect mutations in genes in		C. Kanamycin D. pBR322	III. IV. er fro D – II , D –	resistant genes from a plasmid Cloning vector Bright orange colour bands of DNA m the options given below: II
diagnosis (serum and	using conventional method of lurine analysis), early detection	((4) $A - III, B - II, C - I, C$ Match the columns :	D–I	V
	aked Immunosorbent Assay) is ciple of antigen-antiboody	L	Column I A. Eschereria coli B. Bacillus thuringiens	I.	
How many of the above s (1) Five (3) Three	tatements are correct? (2) Four (4) Two		C. Transgenic mice D. <i>Rosie</i>	III.	Bt toxin Human protein enriched milk
195. The following diagram sh cloning vector pBR322. Fin EcoR ICla I		(Choose the correct answer (1) $A-II, B-I, C-IV, I$ (2) $A-II, B-I, C-III, I$ (3) $A-II, B-III, C-I, I$ (4) $A-IV, B-I, C-II, I$	D–I D–I D–I	V V
Pvu I Pst I pBR. X	BamH I S22 rop]	Match the columns: Column-I A. Elution B. Competence C. Biolistics D. Insertional inactivation	I. II. III.	CERT-XII, Pages 168, 170, 171] Column-II Gene gun Ca^{2+} β -galactosidase gene Extraction of DNA from
	Pvu II ble for recognitions sites rotein involved in replication of	(Choose the correct answer (1) $A-I, B-II, C-IV, I$ (2) $A-III, B-IV, C-I,$ (3) $A-I, B-II, C-III, T$	D – I D – I	II
Plasmid (3) The gene 'X' is resp	onsible for controlling the copy	((4) A-IV, B-II, C-I, I	D–I	II
number of the linked (4) The gene 'X' is r antibiotics.	DNA esponsible for resistance to			asma	its: ati rice was crossed with it was claimed as a novelty
196. Match the columns : Column I	[NCERT-XII, Pages 179, 180, 184] Column II	1	oy an American company Statements II: Tradit	ziona	I knowledge related to
A. Golden RiceB. Bt toxinC. RNAiD. Berin	I. Cry proteinII. Rich in vit. AIII. First transgenic cowIV. Comparison]	_	state	by industrialised nations. [NCERT-XII, Page 185] ements, choose the correct

- С. RNAi Rosie IV. Gene silencing
- D.
- (1) A-II, B-I, C-IV, D-III
- (2) A-IV, B-III, C-I, D-II
- (3) A-II, B-III, C-I, D-IV
- (4) A-IV, B-I, C-II, D-III

(1) Both Statement I and Statement II are true.

answer from the options given below:

- (2) Both Statement I and Statement II are false.
- (3) Statement I is true but Statement II is false.
- (4) Statement I is false but Statement II is true.