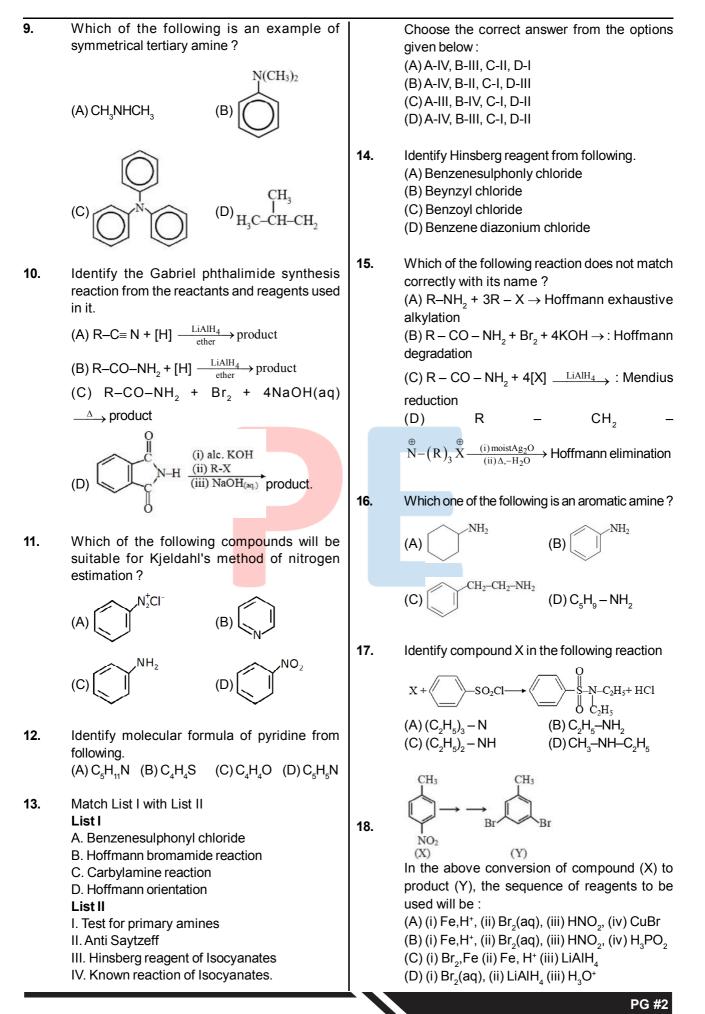
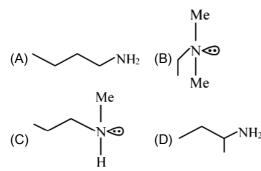


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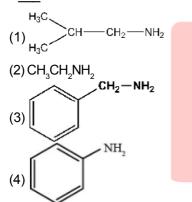
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19. An organic [A] $(C_4H_{11}N)$, Shows optical activity and gives N₂ gas on treatment with HNO₂. The compound [A] reacts with PhSO₂Cl producing a compound which is soluble in KOH. The structure of A is :



- **20.** Given below are two statements : one is labelled as Assertion (A) and the other is labelled as Reason (R) :
- 21. The total number of amines among the following which can be synthesized by Gabriel synthesis is .



22. A reaction of 0.1 mole of Benzylamine with bromomethane gave 23 g Benzyl trimethyl ammonium bromide. The number of moles of bromomethane consumed in this reaction are $n \times 10^{-1}$, when n =_____. (Round off to the Nearest Integer).(Given: Atomic masses: C : 12.0 u, H : 1.0 u, N : 14.0 u,Br : 80.0 u]

23.
$$\underbrace{\overset{\text{NH}_2}{\underset{0-5^{\circ}\text{C}}{\overset{\text{NaNO}_2+\text{HCI}}{\overset{\text{NaNO}_2+\text{HCI}}{\overset{\text{O}-5^{\circ}\text{C}}}}}_{\text{II. NH}_2^+\text{CI}} A \xrightarrow{\underset{1}{\overset{\text{I.}}{\overset{\text{O}-9}{\overset{\text{O}-9}}}}_{\text{II. NH}_2^+\text{CI}} B$$
If molar mas of compound B is x then find $\frac{X}{2}$

24. The total number of reagents from those given below, that can convert nitrobenzene into aniline is (Intergern answer)

I. Sn-HCl	II. Sn-NH₄OH
III. Fe-HCI	IV. Zn-HC1
V. H ₂ -Pd	VI. H ₂ -Raney Nickel

Assertion (A): α – halocarboxylic acid on reaction with dil NH₃ gives good yield of α – amino carboxylic acid whereas the yield of amines is very low when prepared from alkyl halides.

Reason (R): Amino acids exist in zwitter ion form in aqueous medium.

In the light of the above statements, choose the

correct answer form the option given below: (A) (A) is not correct but (R) is correct

(B) Both (A) and (R) correct and (R) is the correct explanation of (A) (A)

(C) Both (A and (R) are correct but (R) is not the correct explanation of (A)

(D) (A) is correct but (R) is not correct

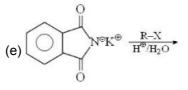
(SECTION-B)

25. The following reactions, how many reactions, are used for the preparation of amines.

(a) R .C = $__{\text{LiAlH}_4}$

(b)
$$\stackrel{O}{\underset{R-C-NH_2}{\overset{\text{LiAlH}_4}{\overset{}}}}$$

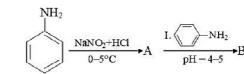
(c) $\stackrel{O}{\underset{R-C-NH_2}{\overset{\otimes}{\underset{Br_2+OH}{\overset{\otimes}{\overset{}}}}}}$



(f)
$$R \xrightarrow{O} CH_3 \xrightarrow{N_2H_4 OH} OH$$

(g)
$$\underset{R-C-NH_2}{\overset{O}{\amalg}} \xrightarrow{P_2O_5}$$

(h) R-CH₂-NO₂
$$\xrightarrow{N_2.Ni}$$



If molar mass of compound B is X then find X.

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26.

PG #3

27. Total number of bromine atom present in major 0 organic product is : NH_2 -H 1.KOH 29. Α 2.Br-CH₂-CH₂-F Major pyridine 1. organic 2. conc.HNO₃+H₂SO₄ product 3. Aq. NaOH 4. H₂/Pd 5. Br₂/H₂O N–H 6. NaNO₂ + HCl (excess) +CТ . N–Н NH, -HN 7. CuBr + HBr28. What will be the net charge on intermediate 0 The molecular weight of the product C will be : formed in the reaction shown below? (N=14, C=12, H=1, Br=79, F=19)) 0 $CH_3 - C - Cl + NH_3 \longrightarrow Intermediate \longrightarrow CH_3 - C - NH_2$ 0 0 $\overset{\parallel}{\mathbb{R}}_{-C-\mathrm{NH}_{2}} + x\mathrm{NaOH} + \mathrm{Br}_{2} \longrightarrow \mathrm{R-NH}_{2} +$ 30. $2NaBr + Na_{2}CO_{3} + H_{2}O$ Number of moles of NaOH used in above Hoffmann bromamide reaction is :