

JEE MAIN : CHAPTER WISE TEST-11

SUBJECT :- CHEMISTRY

DATE.....

CLASS :- 12th

NAME.....

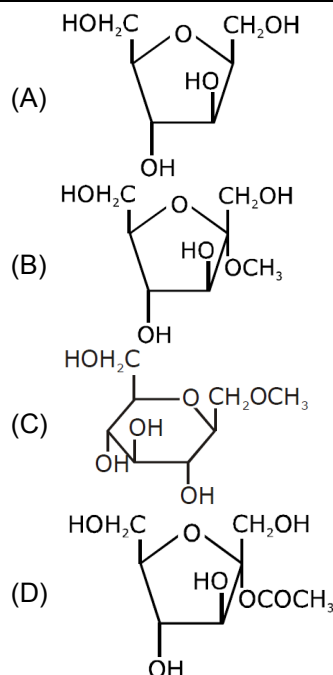
CHAPTER :- BIOMOLECULES

SECTION.....

(SECTION A)

1. Which of the following monosaccharides is a pentose?
(A) Glucose (B) Fructose
(C) Ribose (D) Galactose
2. Ring structure of glucose is due to formation of hemiacetal and ring formation between-
(A) C₁ and C₅ (B) C₁ and C₄
(C) C₁ and C₃ (D) C₂ and C₄
3. Hydrolysis of sucrose is called-
(A) Esterification (B) Saponification
(C) Inversion (D) Hydration
4. A pigment protein in animals is-
(A) Chlorophyll (B) Insulin
(C) Keratin (D) Haemoglobin
5. The main structural feature of proteins is-
(A) An ester linkage
(B) An ether linkage
(C) The peptide linkage
(D) All
6. α -Helical structure refers to the-
(A) Primary structure of protein
(B) Secondary structure of protein
(C) Tertiary structure of protein
(D) Quaternary structure of proteins
7. Simplest amino acid is-
(A) Lysine (B) Glycine
(C) Leucine (D) Alanine
8. Mark the globular protein in the following-
(A) Collagen
(B) Myoglobin or Haemoglobin
(C) Myosin
(D) Fibroin
9. Calorific value is in the order-
(A) Fats > Proteins > Carbohydrates
(B) Carbohydrates > Fats > Protein
(C) Fats > Carbohydrates > Protein
(D) Protein > Fats > Carbohydrates
10. Vitamin necessary for blood clotting is:
(A) E (B) C (C) K (D) D
11. The acidic characters of DNA and RNA are due to:
(A) Purine bases
(B) Sugar molecules
(C) Pyrimidine bases
(D) Phosphoric group
12. According to the Chargaff's law:
(A) A + T = G + C (B) T + A = C
(C) A + T + G = C (D) $\frac{A+G}{C+T} = 1$
13. Which of the following is not a reducing sugar?
(A) Sucrose (B) Galactose
(C) Glucose (D) Lactose
14. The sulphur containing essential amino acids bonded together in long peptide chains form-
(A) Hydrocarbons (B) Nucleic acids
(C) Organic acids (D) Proteins
15. Which of the following substances is involved in all energy transformation reactions in a living system?
(A) Calcium
(B) Phosphate
(C) Cyclic AMP
(D) Creatine phosphate
16. C₂-epimer of D-Glucose is -
(A) D-Glucose (B) D-Allose
(C) D-Altrose (D) D-Mannose
17. C₃-epimer of D-Glucose is -
(A) D-Glucose (B) D-Allose
(C) D-Altrose (D) D-Mannose
18. Which compound can exist in a dipolar (Zwitterion) structure ?
(A) C₆H₅CH₂CH(N=CH₂)COOH
(B) (CH₃)₂CHCH(NH₂)COOH
(C) C₆H₅CONHCH₂COOH
(D) HOOCCH₂CH₂COCOOH

19. Among the following, the **incorrect** statement is :
 (A) Cellulose and amylose has 1,4-glycosidic linkage.
 (B) Lactose contains β -D-galactose and β -D-glucose.
 (C) Maltose and lactose has 1,4-glycosidic linkage.
 (D) Sucrose and amylose has 1,2-glycosidic linkage.
20. Which of the following compounds will behave as a reducing sugar in an aqueous KOH solution ?



(SECTION B)

21. How many stereoisomers are possible by cyclic structure of glucose?
22. How many peptides are negatively charge at pH = 7?
- $\text{NH}_2 - \text{CH}_2 - \text{COOH}$, $\text{Me} - \overset{\text{COOH}}{\underset{|}{\text{CH}}} - \text{NH}_2$,
 $\text{COOH} - (\text{CH}_2)_3 - \overset{\text{COOH}}{\underset{|}{\text{C}}} - \text{NH}_2$,
 $\text{NH}_2 - \text{CH}_2 - \text{CH}_2 - \overset{\text{COOH}}{\underset{|}{\text{CH}}} - \text{NH}_2$,
 $\text{CH}_2 - \text{CH}_2 - \overset{\text{NH}_2}{\underset{\text{HO}}{|}{\text{CH}}} - \text{CH}_2 - \text{COOH}$,
 $\text{HOOC} - \overset{\text{NH}_2}{\underset{|}{\text{CH}}} - \text{CH}_2 - \text{COOH}$,
 $\text{NH}_2 - \text{CH}_2 - \overset{\text{CONH}_2}{\underset{|}{\text{CH}}} - \text{CH}_2 - \text{COOH}$
23. Maximum number of monosaccharide units present in oligosaccharides is
24. Number of chiral carbons in β -D-(+)-glucose is

25. The number of Stereogenic centres in α -D-Glucose are
26. In an amino acid, the carboxylic group ionises at $\text{pK}_{a1} = 2.34$ and ammonium ion at $\text{pK}_{a2} = 9.60$. The isoelectric point of the amino acid is at pH -
27. Consider following carbohydrates and write number of compounds which can not show mutarotation.
 Glucose, Fructose, Cellulose, Starch, Mannose, Galactose, Lactose, Sucrose
28. Number of dipeptide which can be formed by :
 Glycine, Alanine, Leucine, Phenylalanine are
29. What is the total number of basic groups in the following form of lysine ?
 $\text{H}_3\text{N}^+ - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \text{H}_2\text{C} - \overset{\text{O}}{\parallel}{\text{C}} - \text{O}^-$
30. The total number of lone-pairs of electrons in melamine is