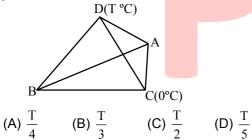
JEE MAIN : CHAPTER SUBJECT :- PHYSICS			DATE	
CLASS :- 11 th		NAME		
СНА	PTER :- THERMAL PHYSICS		SECTION	
	(SECT	ION A)		
1.	Expansion during heating-	á	Two vessels A and B of different materials	
	(A) Occurs only in solids		are similar in shape and size. The same	
	(B) Increases the weight of a material		quantity of ice filled in them melts in times t_1	
	(C) Decreases the density of a material		and t ₂ respectively. The ratio of the thermal conductivities of A and B is -	
	(D) Occurs at the same rate for all solids		(A) $t_1 : t_2$ (B) $t_2 : t_1$	
	and liquids		(C) $t_1^2 : t_2^2$ (D) $t_2^2 ; t_1^2$	
		8.	300 g of water at 25°C is added to 100g of	
2.	Two tanks A and B contains water at 30°C	0.	ice at 0°C. The amount of ice melts is -	
	and 80°C respectively, calculate the		(A) 6.25 g	
	amount of water that must be taken from		(B) 93.75 g	
	each tank to prepare 40kg of water at		(C) 100 g	
	50°C -		(D) none of the above	
	(A) 24 kg, 16 kg (B) 16 kg, 24 kg (C) 20 kg, 20 kg (D) 30 kg, 10 kg			
	$(C) \ge 0$ kg, ≥ 0 kg $(D) \ge 0$ kg, 10 kg			
3. 4.	Mechanism of heat-transfer involved in	9.		
	freezing of lakes in colder region -			
	(A) Conduction only (B) Convection only		Fig.(a) Fig.(b) Fig.(c)	
	(C) Radiation only (D) None of these		Figure (a), (b), (c) shows three different	
	A gas is at pressure P and temperature T.		arrangements of materials 1, 2 and 3 to	
	Coefficient of volume expansion of one		form a wall. Thermal conductivities are K_1	
	mole of gas at constant pre <mark>ssure</mark> is –		> K_2 > K_3 . The left side of the wall is 20°C higher than the right side. Temperature	
	(A) $\frac{1}{T}$ (B) T (C) $\frac{1}{T^2}$ (D) T ²		difference ΔT across the material 1 has	
	(T) T (D) (D) T (D) T^{2} (D) T		following relation, in three cases:	
			(A) $\Delta T_a > \Delta T_b > \Delta T_c$	
5.	A temperature difference of 5°C on Celsius scale corresponding to the		(B) $\Delta T_a = \Delta T_b = \Delta T_c$	
	Celsius scale corresponding to the following temperature difference in the		(C) $\Delta T_a = \Delta T_b > \Delta T_c$	
	Fahrenheit scale -		(D) $\Delta T_a = \Delta T_b < \Delta T_c$	
	(A) 9° (B) 41°	10.	The diameters of steel rods A and B	
	(C) 2.8° (D) 15°		having the same length are 2 cm and 4 cm	
	A solid of mass 2kg is heated and ΔH		respectively. They are heated through	
0.	(Heat given) vs $\Delta\theta$ (change in		100°C. What is the ratio of increase of	
	temperature) is plotted. Specific heat of		length of A to that of B.	
	solid is –		(A) 1 : 2 (B) 2 : 1	
	\uparrow		(C) 1 : 1 (D) 4 : 1	
	\uparrow	-		
		11.	Two liquids are at temperature 20°C and	
	ΔH (in kJ)		40°C. When same mass of both of them is	
	45°		mixed, the temperature of the mixture is	
			32°C. What is the ratio of the their specific	
	(A) 1 J/kg/°C (B) 0.5 J/kg/°C		heats ?	
	(C) 2 kJ/kg/°C (D) 0.5 kJ/kg/°C		(A) 1/3 (B) 2/3 (C) 1/5 (D) 2/5	

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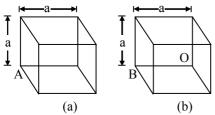
- 12. A steel ball of mass $m_1 = 1$ kg moving with velocity 50 m/sec collides with another steel ball of mass $m_2 = 200$ gm lying on the ground and both come to rest. During the collision their internal energies changes equally and T_1 and T_2 are the rise in temperature of masses m_1 and m_2 respectively. If $s_{steel} = 0.105$ cal/gm °C and J = 4.18, then -(A) $T_1 = 7.1^{\circ}$ C, $T_2 = 1.47^{\circ}$ C (B) $T_1 = 1.42^{\circ}$ C, $T_2 = 7.1^{\circ}$ C (C) $T_1 = 3.4$ K, $T_2 = 17.0$ K
 - (D) None of these
- **13.** If two rods of length L and 2L having coefficient of linear expansion α and 2α respectively are connected so that total length becomes 3L, the average coefficient of linear expansion of the composition rods equals -

(A)
$$\frac{3}{2} \alpha$$
 (B) $\frac{5}{2} \alpha$
(C) $\frac{5}{3} \alpha$ (D) None of these

14. Six similar bars each of thermal resistance R are joined to form a regular tetrahedron as shown in the figure. Point D is maintained at a constant temperature T °C and point C at 0°C. Temperature of junction A is -



- **15.** 10 g of ice at -20° C is added to 10g of water at 50°C. The amount of ice in the mixture at resulting temperature is (Specific heat of ice = 0.5 cal g⁻¹°C⁻¹ and latent heat of ice =80 cal g⁻¹) (A) 10g (B) 5g (C) 0 g (D) 20 g
- **16.** A and B are made up of an isotropic medium. Both A and B are of equal volume. Body B has cavity as shown in Fig.(b). Which of the following statements is true?

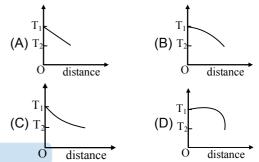


(A) Expansion in volume of A > expansion in B

(B) Expansion in volume of B > expansion in A

(C) Expansion in A = expansion in B (D) None of these

17. The ends of a metal bar of constant crosssectional area are maintained at temperatures T_1 and T_2 which are both higher than the temperature of the surroundings. If the bar is unlagged, which one of the following sketches best represents the variation of temperature with distance along the bar ?



18. 10 gm of ice at 0°C is mixed with 10 gm steam at 100°C in a container of negligible heat capacity. Amount of steam in the mixture after some time will be $(S_w = 1 \text{ cal/gm°C}, L_v = 540 \text{ cal/gm}, L_f = 80 \text{ cal/gm}) - (A) 0 \text{ gm}$ (B) $\frac{20}{3}$ gm

(C) $\frac{10}{3}$ gm (I

(D) None of these

Heat current is maximum in which of the following ? (rods are of identical dimension)
 (A) Cu
 (B) Steel Cu

- $(A) \ \underline{Cu} \qquad (B) \ \underline{Steel} \ \underline{Cu} \qquad (C) \ \underline{Cu} \ \underline{Steel} \qquad (D) \ \underline{Steel} \qquad (D) \ \underline{Steel} \qquad (C) \ \underline{Steel} \qquad (D) \ \underline{Steel} \qquad (C) \ \underline{Steel} \qquad \underline{Steel} \qquad \underline{Steel} \qquad (C) \ \underline{Steel} \ \underline{Steel} \qquad \underline{Steel} \qquad \underline{Steel} \qquad \underline{Steel} \qquad \underline{Steel} \ \underline{Steel} \ \underline{Steel} \ \underline{Steel} \qquad \underline{Steel} \ \underline{Ste$
- 20. A cylinder of radius R, made of a material of thermal conductivity k₁, is surrounded by a cylindrical shell of inner radius R and outer radius 2R. The shell is made of a material of thermal conductivity k₂. The two ends of the combined system are maintained at two different temperatures. There is no loss of heat across the cylindrical surface and the system is in steady state. The effective thermal conductivity of the system is -

(A)
$$k_1 + k_2$$
 (B) $\frac{k_1 k_2}{k_1 + k_2}$

(C)
$$\frac{k_1 + 3k_2}{4}$$
 (D) $\frac{3k_1 + k_2}{4}$

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(SECTION B)

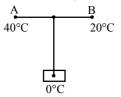
26.

- 21. A thermometer of mass 50 gm and specific heat 0.4 cal/gm/°C reads 10°C. It is then inserted into 1 kg of water and reads 40°C in thermal equilibrium. The temperature of water before insertion of thermometer in 10 °C is (Neglect other heat losses).
- A bar measured with a Vernier Caliper is found to be 1800 mm long. The temperature during the measurement is 10°C.
 The measurement error if the scale of the

Vernier Caliper has been graduated at a temperature of 20°C is found to $x \times 10^{-2}$ mm. Find x.

- 23. A calorimeter of negligible heat capacity contains 100 gm water at 40°C. The water cools to 35°C in 5 min.. If water is now replaced by a liquid of same volume as that of water at same initial temperature it cools to 35°C in 2 min. Given sp. heats of water and liquid are 4200 J/kg-°C and 2100 J/kg-°C respectively. Find the density of liquid give answer in...× 100 kg/m³. [Assume Newton Iaw of cooling is applicable]
- 24. In the figure shown AB is a rod of length L and thermal resistance $R_H = 10$ SI unit and end A and B are maintained by 20°C and 40°C. At mid point of rod another rod of thermal resistance $R'_H = R_H/4$ SI unit connected and other end of rod is inserted into ice.

After steady state reach, we start counting of time, and find the amount of ice (in kg) melt in 5.6 × 10^4 s. (L_f = 3.36×10^5 J/kg)



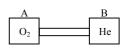
25. Two rods of same dimensions, but made of different materials are joined end to end with their free ends being maintained at 100°C and 0°C respectively. The temperature of the junction is 70°C. Then the temperature of the junction if the rods are interchanged will be equal to T °C Find T :

Two containers A and B are connected by a conducting solid cylindrical rod of length

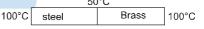
 $\frac{242}{7}$ cm and radius $\sqrt{8.3}$ cm. Thermal

conductivity of the rod is 693 watt/mole-K. The container A contains two mole of oxygen gas and the container B contains four mole of helium gas. At time t = 0 temperature difference of the containers is 50°C, after what time (in seconds) temperature difference between them will be 25°C. Transfer of heat takes place through the rod only. Neglect radiation

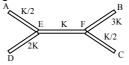
loss. Take R = 8.3 J/mole-K and $\pi = \frac{22}{\pi}$.



- 27. 1 gm of steam at 100°C is kept in a vessel of negligible heat capacity. Amount of ice required at 0°C so that at equilibrium only water is remaining at 0°C is m gm. Find m.
- **28.** Figure shows a steel rod joined to a brass rod. Each of the rods has length of 31 cm and area of cross-section 0.20 cm². The junction is maintained at a constant temperature 50°C and the two ends are maintained at 100°C. The amount of heat taken out from the cold junction in 10 minutes after the steady state is reached in n × 10² J. Find 'n'. The thermal conductivities are K_{steel} = 46 W/m–°C and K_{brass} = 109 W/m–°C. $_{50°C}$



29. Five rods with identical geometries are arranged as shown. Their thermal conductivity are shown. Only A and C are maintained at 100°C and 0°C respectively. If temperature difference between ends B and C can be writted as 10x °C where x is an integer. Then find x.



30. One end of copper rod of uniform crosssection and of length 1.45 m is in contact with ice at 0°C and the other end with water at 100°C. The position of point along its length where a temperature of 200°C should be maintained so that in steady state the mass of ice melting is equal to that of steam produced in the same interval of time is x cm from hotter end of rod. Find x [Assume that the whole system is insulated from surroundings]. (take $L_v =$ 540 cal/g $L_f = 80$ cal/g)