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		NEET : CHAPTE	R WISE	TEST-12	
			DATE		
	CLASS :- 11 <sup>th</sup>			NAME	
CHA	PTER :- NOMENCLATUR		FION-A)	SECTION	
1.	Compound having oper		9.	IUPAC name of compound	
	(A) Pentane	(B) Isopentane		$HO - C = O CH_3$	
	(C) Neopentane	(D) All the above			
				$CH_3 - C = C - C - H$ is:	
2.	Which is an acyclic com	npound?		 NH <sub>2</sub> Cl	
	(A) Methane	(B) Benzene		(A) 2-Amino-3-chloro-2-methylpent-2-enoic	
	(C) Pyrrole	(D) Cyclobutane		acid	
3.	The total number of secondary H-atoms in			(B) 3-Amino-4-chloro-2-methylpent-2-enoic acid	
	the structure given belo	w are :		(C) 4-Amino-3-chloro-2-methylpent-2-enoic	
	$(CH_3)_2CHCH_2C_2H_5$			acid	
	(A) 1 (B) 4	(C) 3 (D) 2		(D) All of the above	
4.	Iso-octane contains		10.	Identify the compound which is	
	(A) five 1°, one 2° and two 3° carbon			homocyclic, aromatic, and unsaturated?	
	atoms			$\frown$	
	<ul> <li>(B) four 1°, two 2° and one 3° carbon atoms</li> <li>(C) five 1°, one 2°, one 3° &amp; one 4° carbon</li> </ul>			(A) (B)	
				$\frown$	
	atoms			(C) (D)	
	(D) five 1°, two 2°, one	3° & one 4° carbon		чŪг	
	atoms		11.	The correct name of 3,3-dimethyl	
5.	How many 1°carbon at	oms <mark> will b</mark> e present		propanamide is	
	in the simplest hydroca	arbon <mark> hav</mark> ing two 3°		<ul><li>(A) 2-Methyl butanamide</li><li>(B) 3-Methyl butanamide</li></ul>	
	and one 2° carbon aton	ns?		(C) Iso-propyl ethanamide	
	(A) 3 (B) 4	(C) 5 (D) 6		(D) Iso propyl acetamide	
6.	How many carbons a	re in the simplest	12.	The IUPAC name of	
	alkyne having two side			$CH_3 - CH_2 - CH - COOC_2H_5$	
	(A) 5 (B) 6 (C) 7 (D) 8			is	
				$CH_3$	
7.	Which of the following is the first member			<ul><li>(A) 2-Ethyl-ethyl acetate</li><li>(B) Ethyl 3-methyl butanoate</li></ul>	
	of ester homologous se	ries?		(C) Ethyl 2-methyl butanoate	
	(A) Ethyl ethanoate			(D) 2-Methyl butanoic acid ethyl ester	
	<ul><li>(B) Methyl ethanoate</li><li>(C) Methyl methanoate</li></ul>		13.	The IUPAC name of	
	(D) Ethyl methanoate		15.	CH <sub>2</sub> – CHO	
				-	
8.	The IUPAC name of is -			$OHC - CH_2 - CH_2 - CH - CH_2 - CHO$ is:	
0.		Sr Br		(A) 4, 4-Di(formylmethyl) butanal	
	(A) 2-Bromo-4-isopropy	•		(B) 2-(Formylmethyl) butane-1, 4- dicarbaldehyde	
	<ul> <li>(B) 2, 3-Dimethyl-5-broi</li> <li>(C) 2-Bromo-4, 5-dimet</li> </ul>			(C) Hexane-3-acetal-1, 6-dial	
	(D) 5-Bromo-2, 3-dimet	•		(D) 3-(Formylmethyl) hexane-1, 6-dial	
			•		

14. Which of the following is crotonic acid? (A)  $CH_2 = CH - COOH$ (B) C<sub>6</sub>H<sub>5</sub>–CH=CH–COOH (C) CH<sub>3</sub>-CH=CH-COOH CH-COOH (D) 📗 CH – COOH  $CH_3$ The derived name of  $(CH_3)_3$  C – C – OH is – 15.  $\dot{C}_2H_5$ (A) t-Butyl ethyl methyl carbinol (B) t-Butyl ethylethanol (C) t-Butyl ethyl methyl methanol (D) None of these  $CH_3-CH(CH_3)CH_2-C \equiv C-CH=CH_2$ 16. its derived name is (A) 6-Methyl-1-heptenyne-3 (B) Iso-butyl vinyl acetylene (C) Iso hexynyl ethylene (D) None 17. Which of the following is not correctly matched? (A) Acetonitrile  $CH_2 = CHCN$ (B) Allyl chloride  $CH_2 = CH - CH_2CI$ (C) s-Butyl group  $CH_3 - CH_2 - C_2H_5$ (D) Ethylene dichloride CH<sub>2</sub>Cl –CH<sub>2</sub>Cl 18. IUPAC name of,  $OHC - CH = CH - CH - CH = CH_{2}$ is : CH,CH,CH,CH, (A) 4-Butyl-2,5-hexadien-1-al (B) 5-Vinyloct-3-en-1-al (C) 5-Vinyloct-5-en-8-al (D) 3-Butyl-1, 4-hexadien-6-al 19. The IUPAC name of  $CH_3 - CH - CH_2 - C(CH_3)_2$  is: OH OH (A) 2-Methyl-2, 4-dihydroxy propane (B) 2,2-Dimethyl-4-hydroxy butanol (C) 2-Methyl-2, 4-pentane diol (D) 2-Hydroxy-4, 4-dimethyl butanol-4 20. The IUPAC name of (CH<sub>3</sub>)<sub>2</sub> CH-CO-O-CH<sub>2</sub>-CH<sub>3</sub> is : (A) Ethyl butyrate (B) Isopropyl propionate (C) Ethyl-2-methyl propanoate (D) Isobutyl ethanoate 21. One among the following is the correct IUPAC name for the compound Н CH<sub>3</sub>CH<sub>2</sub> - N - CHO (A) N-Formyl aminoethane (B) N-Ethyl formyl amine (C) N-Ethyl methanamide (D) Ethylamino methanal.

22. The systematic name for  $HO - C - C = C - CH - CH_3$ is : O CH<sub>3</sub> NH<sub>2</sub>Cl (A) 2-Methyl-3-amino-4-chloro-2-pentenoic acid (B) 1-Hydroxy-1-oxo-2-methyl-3-amino-4chloro-2- pentene (C) 3-Amino-4-chloro-2-methyl-2-pentenoic acid 3-Amino-2, 4-dimethyl-4-chloro-2-(D) butenoic acid  $CH_3 - O - C - CH_2 - COOH$ 23. 0 The correct systematic name of the above compound is : (A) 2-Acetoxy ethanoic acid (B) 2-Methoxy carbonyl ethanoic acid (C) 3-Methoxy formyl ethanoic acid (D) 2-Methoxy formyl acetic acid 24. The IUPAC name of  $CH_2 = CH$ Н  $HC \equiv C - CH_2 - CH_2 - C = CH - C = O$  is: (A) 3-(1-Butynyl)-3-vinyl-2-pentadienal (B) 5-Ethynyl-3-vinyl-2-pentenal (C) 3-Vinyl-2-heptene-6-ynal (D) 5-Acetylenyl-3-ethenyl-2-pentenal 25. The IUPAC name of 0  $\|$  $(CH_3)$ ,  $CH(CH_2)$ ,  $-C - N(CH_3)$ , is -(A) N, N, 4-Trimethylpentanamide (B) Dimethylamino-4-methylpentanone (C) N, N-Dimethylamino-4-methylpentanamide (D) 2-Methyl-5-oxodimethylpentanamine The IUPAC name of 26. is – ĠН (A) 5-Hydroxy-3, 4-hxanedione (B) 3, 4-Dioxo-2-hexanol (C) 2-Hydroxy-3, 4-hexanedione (D) 2-Hydroxy-3, 4-diketohexane The IUPAC name of 27. - is -(A) 3-Cyclobutylcyclopropane (B) 1-Cyclopropyl-3-methylcyclobutane (C) 1, 3-Cyclopropylmethylcyclobutane (D) 2-Methylbicyclo [4.3.0] heptane  $C \equiv CH$ HO, 28. is -The IUPAC name of (A) 1-Ethynyl-1-hydroxycyclohexane (B) 1-(Hydroxycyclohexyl) ethyne (C) 1-Ethynylcyclohexanol

(D) 1-Acetylenyl-1-hydroxycyclohexane

29.	The IUPAC name of COOH (A) 6-Carboxy-2, 3-dimethyl-4-heptenoic acid (B) 2, 3, 6-Trimethyl-1-heptene-1, 7-dioic acid (C) 2, 5, 6-Trimethyl-3-heptenedioic acid (D) 6-Carboxy-2, 5-dimethyl-3-heptenoic acid
30.	The IUPAC name of [(CH <sub>3</sub> ) <sub>3</sub> C] <sub>4</sub> C is - (A) Tetra-t-butyl methane (B) 3, 3-Bis(1, 1-dimethyl ethyl) -2, 2, 4, 4- tetramethyl pentane (C) Tetra kis (1, 1-diemthyl ethyl) methane (D) Tetraneobutyl methane
31.	Which of the following are secondary radicals? (a) $CH_3 - \dot{C}H - C_2H_5$ (b) $CH_2 = \dot{C} - CH_3$ (c) $CH_2 = CH \cdot$ (d) $(CH_3)_2CH \cdot$ (A) a, b, c (B) a, d, c (C) b, c, d (D) a, b, d
32.	Common name of the structure   CH <sub>2</sub> – OH (A) Ethylene Glycol (B) Ethene dialcohol (C) Glycerol (D) Ethylene alcohol
33.	$\begin{array}{cccc} Common & name & of & the & compound \\ & & \\ & & \\ & & \\ CH_3-CH_2-C-NH_2 & \\ \hline \\ (A) & Acetamide & (B) & Propionamide \\ (C) & Butyramide & (D) & Acetic & amide \\ \end{array}$
34.	The correct IUPAC name of the spiro compound is – (A) 1-Oxaspiro [4.3] octane (B) 5-Oxaspiro [3.4] octane (C) 1-Oxaspiro [3.4] octane (D) 5-Oxaspiro [4.3] octane
35.	The IUPAC name of the spiro compound $\swarrow C_2H_5$ is (A) 8-Ethyl [4.5] deca-1, 7-diene (B) 2-Ethyl spiro [5.4] deca-1,6-diene (C) 3-Ethyl spiro [5.4] deca-3,7 diene (D) 2-Ethyl spiro [4.5] deca-2,6-diene

(SECTION-B) Number of 4° carbon atoms present in the 36. compound are :-CH<sub>3</sub> CH<sub>3</sub>  $CH_3 - C - CH - CH_2 - CH_3$ CH<sub>3</sub> (A) 4 (B) 2 (C) 3 (D) 1 37. 3-Methyl-2-pentanone is : O CH<sub>3</sub> (A)  $\begin{array}{c} & \parallel & \uparrow \\ CH_3 - C - CH - CH_2 - CH_3 \end{array}$ CH<sub>3</sub>-CH-CH<sub>2</sub>-COOH (B)  $\dot{C}H_{3}$  $CH_3$ (C)  $C_2H_5-CH=CH_2-CH_3$ (D)  $CH_3$ — $CH_2$ — $CH = CH_2$ 38. The IUPAC name for the compound  $\begin{array}{c} CI \\ H_3C \end{array} > C = C < \begin{array}{c} CH_2CH_3 \\ I \end{array}$ (A) 3-lodo-4-chloro-3-pentene (B) 2-Chloro-2-iodo-2-pentene (C) 2-Chloro-3-iodo-2-pentene (D) 3-lodo-4-chloro-2-pentene 39. The correct decreasing order of priority for the functional groups of organic compounds in the IUPAC system of nomenclature is (A) -SO<sub>3</sub>H, -COOH, -CONH<sub>2</sub>, -CHO (B) – CHO, – COOH, – SO<sub>3</sub>H, – CONH<sub>2</sub> (C) –CONH<sub>2</sub>, –CHO, –SO<sub>3</sub>H, –COOH (D) -COOH, -SO<sub>3</sub>H, -CONH<sub>2</sub>, -CHO 40. Of the following compounds which has a wrong IUPAC name? I.  $CH_3 - CH_2 - CH_2 - CH_2 - O - CH_2 - CH_3$ Ethyl butanoate  $\begin{array}{ccccccc} \mathrm{II.} \ \mathrm{CH}_3 - \ \mathrm{CH} \ - \ \mathrm{C} - \ \mathrm{CH} \ - \ \mathrm{CH}_3 \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\$ 2,4 Dimethyl -3- pentanone CH<sub>3</sub>  $III. H_{3}C - CH - CH - CH_{3}$ ÓН 2-Methyl -3- butanol

IV.  $CH_3 - CH_2 - CH_2 - CHO$ Butanal (A) I (B) II (C) III (D) IV

PG #3

41. The IUPAC name of CH<sub>3</sub>COCH(CH<sub>3</sub>)<sub>2</sub> is -Which of the following acids does not 46. (A) 4-Methylisopropyl ketone exhibit optical isomerism ? (B) 3-Methyl-2-butanone (A) Maleic acid (B) α-amino acids (C) Lactic acid (D) Tartaric acid (C) Isopropylmethyl ketone (D) 2-Methyl-3-butanone 47. The correct statement regarding the comparison of staggered and eclipsed 42. Which nomenclature is not according to conformations of ethane, is: IUPAC system ? (A) The staggered conformation of ethane  $CH_3 - CH - CH - CH_2CH_3$ is more stable than eclipsed conformation, CH because staggered conformation has no (A) torsional strain. (B) The staggered conformation of ethane 2-Methyl-3-phenylpentane is less stable than eclipsed conformation, CH<sub>3</sub>-C-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>COOH because staggered conformation has (B) Ö torsional strain. (C) The eclipsed conformation of ethane is 5-Oxohexanoic acid more stable than staggered conformation, Br-CH<sub>2</sub>-CH=CH<sub>2</sub> (C) because eclipsed conformation has no 1-Bromo-prop-2-ene torsional strain. CH<sub>3</sub> (D) The eclipsed conformation of ethane is CH<sub>3</sub>-CH<sub>2</sub>-CH-CH<sub>2</sub>-CHCH<sub>3</sub> more stable than staggered conformation (D) even though the eclipsed conformation **b**r ĊH, has torsional strain. 4-Bromo,2,4-di methylhexane 48. Which of the following biphenyls is optically active 43. Structure of the compound whose IUPAC name is 3-Ethyl-2-hydroxy-4-methylhex-3en-5-ynoic acid is -COOH СООН 49. In which of the following molecules, all atoms are coplanar? 44. The total number of  $\pi$  -bond electrons in the followng structure is -(B) H.C (D) (C) (A) 8 (B) 12 (C) 16 (D) 4 45. The IUPAC name of the compound 50. With respect to the conformers of ethane, which of the following statements is true? (A) Bond angle remains same but bond length changes (B) Bond angle changes but bond length

(A) 5-Formylhex-2-en-3-one

- (B) 5-Methyl-4-xoohex-2-en-5-al
- (C) 3-Keto-2-methylhex-5-enal
- (D) 2-Methyl-3-oxopent-5-enal

changes (D) Both bond angles and bond length remains same

(C) Both bond angle and bond length

remains same