

Class: XIIth

Date:

**Solutions** 

Subject: PHYSICS DPP No.: 2

## Topic :-.NUCLEI

1 **(b)** 

 $\beta$  – decay from nuclei is based on this process only

2 **(c**)

The binding energy of nucleus may be defined as the energy equivalent to the mass defect of the nucleus.

If  $\Delta m$  is mass defect than according to Einstein's mass energy relation.

**Binding Energy** 

$$= \Delta mc^{2} = [\{Zm_{p} + (A - Z)m_{n}] - M]c^{2}$$

$$= (7 \times 1.00783 + 7 \times 1.00867 - 14.00307)c^{2}$$

or BE = 
$$0.1124 \times 931.5 \,MeV$$

or 
$$BE = 104.6$$

3 **(a**)

Ionisation energy of  $Li^{++} = 9hcR$ 

Ionization energy = 
$$RchZ^2 = Rch(3)^2$$
 (as  $Z = 3$  for  $Li^{++}$ )

$$=9hcR$$

4 **(b)** 

$$E_b + E_c > E_a$$

5 **(b**)

$$r = \frac{n^2}{Z}(r_0); \Rightarrow r_{(n=2)} = \frac{(2)^2}{2} \times 0.53 = 1.06 \text{ Å}$$

6 **(b)** 

Linear momentum = 
$$mv = 9.1 \times 10^{-31} \times 2.2 \times 10^6$$
  
=  $2.0 \times 10^{-24} kg - m/s$ 

7 **(c)** 

According to the quark model, it is possible to build all hadrons using 3 quarks and 3 antiquarks

Mesons and baryons are collectively known as hadrons

8 **(a)** 

N = M - Z = Total no. of nucleons – no. of protons

10 **(a)** 

Nuclear density is constant hence, mass  $\propto$  volume

Or  $m \propto V$ 

11 **(c)** 

 $_{92}U^{235}$  is normally fissionable

13 **(c**)

Out side the nucleus, neutron is unstable (life -932 s)

14 **(a**)

The mass of nucleus formed is always less than the sum of the masses of the constituent protons and neutrons *i.e.*,  $m < (A - Z)m_n + Zm_p$ 

15 **(c**)

Binding energy per nucleon increases with atomic number. The greater the binding energy per nucleon the more stable is the nucleus

For  $_{26}Fe^{56}$  number of nucleons is 56

This is most stable nucleus, since maximum energy is needed to pull a nucleon away from it

16 **(a)** 

$$X(n,\alpha) {}_{3}^{7}Li \Rightarrow {}_{Z}X^{A} + {}_{0}n^{1} \rightarrow 3^{Li^{7}} + {}_{2}He^{4}$$
  
 $Z = 3 + 2 = 5 \text{ and } A = 7 + 4 - 1 = 10$   
 $\therefore {}_{5}X^{10} = {}_{5}B^{10}$ 

17 **(a)** 

$$\frac{N}{N_0} = \left(\frac{1}{2}\right)^n \Rightarrow \frac{1}{8} = \left(\frac{1}{2}\right)^n \Rightarrow n = 3$$

Now 
$$t = n \times T_{1/2} = 3 \times 3.8 = 11.4 \ days$$

18 **(c**)

Experimental measurements show that volume of a nucleus is proportional to its mass number *A*. If *R* is the radius of the nucleus assumed to be spherical, then its volume

$$\left(\frac{4}{3}\pi R^3\right) \propto A$$

$$R \propto A^{1/3}$$

$$R = R_0 A^{1/3}$$

where  $R_0$  is an empirical constant whose value is found to be 1.1  $\times$  10<sup>-15</sup> m.

20 **(a**)

Rest energy of an electron  $= m_e c^2$ 

Here  $m_e = 9.1 \times 10^{-31} kg$  and c = velocity of light

$$\therefore$$
 Rest energy =  $9.1 \times 10^{-31} \times (3 \times 10^8)^2$  joule

$$= \frac{9.1 \times 10^{-31} \times (3 \times 10^8)^2}{1.6 \times 10^{-19}} eV = 510 \text{ keV}$$

ANSWER-KEY										
Q.	1	2	3	4	5	6	7	8	9	10
A.	В	С	A	В	В	В	С	A	В	A
Q.	11	12	13	14	15	16	17	18	19	20
A.	С	A	С	A	С	A	A	С	A	A

