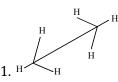


CLASS: XIIth DATE:

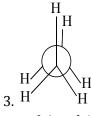
SUBJECT: CHEMISTRY

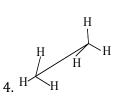
DPP NO.: 5

1. In the following structures which two forms are staggered conformation of ethane?









- a) 1 and 4
- b) 2 and 3
- c) 1 and 2
- d) 1 and 3

2. A mixture of ethane, ethene and ethyne is passed through ammoniacal AgNO₃ solution. The gases which remain unreacted are:

- - a) Ethane and ethene b) Ethane and ethyne c) Ethene and ethyne d) Ethane only

3. In the reaction,

 $C_6H_5CH_3 \xrightarrow{Oxidation} A \xrightarrow{NaOH} B \xrightarrow{Soda lime} C$

The product *C* is

- a) C_6H_5OH
- b) C_6H_6
- c) C₆H₅COONa
- d) C₆H₅ONa

 $A_{(II)H_2O_2,OH^-}^{(I)BH_3.THF}CH_3C \equiv CH_{H_2SO_4}^{HgSO_4}B$

Identify A and B

a) CH₃CHO, CH₃COCH₃

b) CH₃CH₂CHO, CH₃COCH₃

c) CH₃CH₂CHO, CH₃COCH₂CH₃

d) HCHO, CH₃COCH₃

- 5. Cyclobutadiene is said to be
 - a) aromatic
- b) aliphatic
- c) non-aromatic
- d) None of these

- 6. Acetylene reacts with hypochlorous acid to form
 - a) Cl₂CHCHO
- b) ClCH₂COOH
- c) Cl₃COCl
- d) ClCH2CHO

- 7. To enable easy detection of gas leakage from cylinders, the substance added to L.P.G. is:
 - a) Glycols
- b) Phenols
- c) Thioalcohols
- d) Glycerols
- 8. Octane no. of 2,3,3-trimethylbutane has been assumed to be:
 - a) 100
- b) -45
- c) 124

d) Zero

- 9. C_4H_6 may contain
 - a) One double bond
- b) Two double bond
- c) One triple bond
- d) Both (b) and (c)
- 10. Which of the following compounds can form metallic derivatives?
 - a) Ethane
- b) Propyne
- c) 2-butyne
- d) 2-butene
- 11. Increasing order of volatility of C₂H₆, C₂H₄, C₂H₂ and C₆H₆ is:
 - a) C_6H_6 , C_2H_6 , C_2H_4 , C_2H_2 b) C_2H_2 , C_2H_4 , C_2H_6 , C_6H_6 c) C_6H_6 , C_2H_2 , C_2H_4 , C_2H_6 d) C_2H_2 , C_2H_6 , C_2H_4 , C_6H_6
- 12. Octane no. of a fuel can be increased by:
 - a) Isomerism
- b) Alkylation
- c) Reforming
- d) All of these

- 13. 1-propanol on dehydration with H₂SO₄ produces:
 - a) $CH_3 CH = CH_2$
 - b) $CH_3 CH = CH CH_3$
 - c) CH₃CH₂CH₂OCH₂CH₂CH₃
 - d) $CH_3CH_2CH_2CH_2CH = CH_2$



- 14. Propadiene, C₃H₄ molecule contains:
 - a) Two sp^2 and one sp-hybrid carbon
 - b) One sp^2 and two sp-hybrid carbons
 - c) One sp^2 and three sp-hybrid carbons
 - d) None of the above
- 15. Catalyst used in dimerization of acetylene to prepare chloroprene is:
 - a) $HgSO_4 + H_2SO_4$
- b) Cu₂Cl₂
- c) $Cu_2Cl_2 + NH_4Cl$
- d) $Cu_2Cl_2 + NH_4OH$
- 16. Cyclopentene on treatment with alkaline KMnO₄ gives:
 - a) Cyclopentanol
 - b) *trans*-1,2-cyclopentanediol
 - c) cis-1,2-cyclopentanediol
 - d) 1 : 1 mixture of *cis*-and *trans*-1,2-cyclopentanediol
- 17. $C_7H_8 \xrightarrow{3Cl_2, \text{ Heat}} A \xrightarrow{Fe/Br_2} B \xrightarrow{Zn/HCl} C$

Here, the compound *C* is

- a) 3-bromo 2,4,5,6-trichlorotoluene
- b) o-bromo toluene

c) *p*-bromo toluene

d)*m*-bromo toluene

- 18. Naphalene is an example of
 - a) Polynuclear hydrocarbon
 - c) heterocyclic compound

- b) alicyclic compound
- d) aliphatic compound
- 19. Which of the following will give *trans*-diols?

$$c = c < \frac{1. \text{ KMnO}_4}{2. \text{ H}_2\text{O}}$$

$$\frac{1. \text{ OsO}_4, 25^{\circ}\text{C}}{2. \text{ Na}_2\text{SO}_3}$$

- $c = c < \frac{1. \text{ OsO}_4}{2. \text{ Na}_2 \text{SO}_3}$
- $\frac{35\% \text{ H}_2\text{O}_2}{\text{HCO}_2\text{H}, 25^{\circ}\text{C}}$

- 20. Benzene can react with
 - a) Bromine water
- b) HNO₃
- c) H₂O
- d) CH₃OH

