

CLASS: XIIth DATE:

SUBJECT: CHEMISTRY

DPP NO.: 2

Topic:-HYDROCARBONS

1. A mixture of nitrogen and acetylene, on passing electric spark through it gives:

a) Hydrogen and carbon b)

Hydrogen cyanide

c) Nitromethane

d) Nitroethane

2. In the sequence of reactions,

$$C_2H_4 \xrightarrow{\text{HBr}} X \xrightarrow{\text{AgCN}} Y \xrightarrow{\text{(H)}} Z$$
,

Compound Zis

a) N-methyl ethanamine

b) N-propylamine

c) N, N-dimethylamine

d) Ethyl cyanide

3. Which one of these is not true for benzene?

a) It forms only one type of monosubstituted product.

b) There are three carbon-carbon single bonds and three carbon-carbon double bonds

c) The heat of hydrogenation of benzene is less than the theoretical value.

d) The bond angle between the carbon-carbon bonds is 120°.

4. Presence of a nitro group in a benzene ring

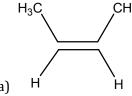
a) Activates the ring towards electrophilic substitution

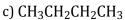
b) Renders the ring basic

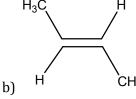
c) Deactivates the ring towards nucleophilic substitution

d) Deactivates the ring towards electrophilic substitution

5. The major product in the reaction of 2-butyne with Li/liq. NH_3 is







$$d) H_2 C = CH - CH_2 - CH_3$$

6. Hydrocarbon liquid at STP is:

- a) Ethane
- b) Propane
- c) Butane

d) Pentane

7. Chlorination of benzene is not possible in the following reaction

a)
$$C_6H_6 + Cl_2 \xrightarrow{FeCl_3}$$

b)
$$C_6H_6 + HOCl \xrightarrow{H^+}$$

b)
$$C_6H_6 + HOCl \xrightarrow{H^+}$$
 c) $C_6H_6 + I - Cl \xrightarrow{ZnCl_2}$ d) $C_6H_6 + Cl_2 \xrightarrow{AlCl_3}$

$$d) C_6 H_6 + Cl_2 \xrightarrow{AlCl_3}$$

- 8. In the series, ethane, ethene and ethyne, the C-H bond energy is
 - a) Same in all the three compounds
- b) Greatest in ethane

c) Greatest in ethene

d) Greatest in ethyne

- 9. The shape of 2-butene is:
 - a) Linear
- b) Planar
- c) Tetrahedral
- d) Pyramidal

- 10. The substance used as an anti-knock compound is:
 - a) Tetraethyl lead
- b) Lead tetrachloride
- c) Lead acetate
- d) Ethyl acetate

- 11. Petroleum refining is:
 - a) Obtaining aromatic compounds from aliphatic compounds in petroleum b)
- Cracking of petroleum to get gaseous hydrocarbons

c) Purification of

- petroleum
- d) Distillation of petroleum to get different fractions
- 12. Zinc-copper couple that can be used as a reducing agent is obtained by:
 - a) Mixing zinc dust and copper gauze
 - b) Zinc coated with copper
 - c) Copper coated with zinc
 - d) Zinc and copper wires welded together
- 13. Which of the following hydrocarbons has the lowest dipole moment?

$$\begin{array}{c} \text{H}_{3}\text{C} \\ \text{a)} \quad \text{H} \end{array}$$

$$\begin{array}{ccc} & & & \\ &$$

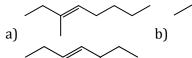
b)
$$CH_3C \equiv CCH_3$$

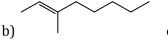
c)
$$CH_3CH_2C \equiv CH$$

d)
$$CH_2 = CH - C \equiv CH$$

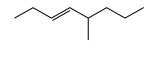
- 14. Alkene

d)









- 15. A solution of sodium salt of fatty acid was electrolysed during Kolbe's reaction. The solution left after electrolysis is:
 - a) Richer in NaOH
- b) Richer in H₂SO₄
- c) Richer in sodium salt d) All of these

16. Sample of 2,3-dibromo-3-methylpentane is heated with zinc dust. The resulting product is isolated and heated with HI in the presence of phosphorus. Indicate which is the structure that represents the final organic product in the reaction?

$$CH_{2} = CH - CH - CH_{2} - CH_{3}$$

$$| CH_{3} = CH_{2} - CH - CH_{2} - CH_{3}$$

$$| CH_{3} = CH_{2} - CH - CH_{2} - CH_{3}$$

$$| CH_{3} = CH - CH - CH_{2} - CH_{3}$$

$$| CH_{3} = CH - CH_{3} - CH_{3}$$

$$| CH_{2} = CH - CH - CH_{2} - CH_{3}$$

$$| CH_{3} = CH_{3} - CH_{3}$$

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$$| CH_{3} = CH_{3} - CH_{3}$$

17. Which compound does not give precipitate with ammoniacal silver nitrate solution?

a)
$$C_2H_5 - C \equiv CH$$

c) CH_3
 $|$ $CH_3 - CH - C \equiv CH$

b)
$$CH_3 - C \equiv C - CH_3$$

d)
$$Ph - CH_2 - C \equiv CH$$

- 18. Hydroxylation of propyne in the presence of HgSO₄/H₂SO₄ is initiated by the attack of:
 - a) Carbene

d)

- b) Free radical
- c) Electrophile
- d) Nucleophile
- 19. Benzene vapour mixed with air when passed over V₂O₅ catalyst at 775 K gives
 - a) Glyoxal
- b) Oxalic acid
- c) Maleic anhydride
- d) Fumaric acid
- 20. Kolbe's synthesis on electrolysis of sodium salt of butanoic acid gives :
 - a) *n*-hexane
- b) Isobutene
- c) Butane
- d) Ethene