

**CLASS: XIIth** DATE:

**SUBJECT: CHEMISTRY** 

**DPP NO.: 8** 

1.	On mixing a certain alkane with chlorine and irradiating it with ultraviolet light, it forms only
one	monochloroalkane

- a) Propane
- b) Pentane
- c) *Iso*-pentane
- d) Neo-pentane
- 2. Formation of alkane by the action of Zn on alkyl halide is called:
- a) Wurtz reaction
- b) Kolbe's reaction
- c) Cannizzaro's reaction d)
- Frankland's

reaction

- 3. Chloretone used as a drug is prepared by the reaction of acetone with:
  - a) Chlorine
- b) Ethyl chloride
- c) Chloroform
- d) Ethylene dichloride

- 4. Which is gem dihalide?
  - a)  $CH_3 \cdot CHBr_2$
- b)  $CH_2Br \cdot CH_2Br$
- c)  $CH_3 \cdot CHBr \cdot CH_2Br$  d) None of these

- 5. Which of the following is a Grignard reagent?
  - a) Ammoniacal solution of AgNO<sub>3</sub>
  - b) Ethereal solution of C<sub>2</sub>H<sub>5</sub>MgCl
  - c) Alcoholic solution of KOH
  - d) Aqueous solution of caustic soda

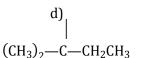


- 6. The product formed on reaction of ethyl alcohol with bleaching powder is
  - a) CHCl<sub>3</sub>b)
- $CCl_3CHOc)$
- $CH_3COCH_3d$ )
- CH<sub>3</sub>CHO

- 7. Chloral is:
  - a) CCl<sub>3</sub>CHO
- b)  $CCl_3 \cdot CO \cdot CH_3$
- c)  $CCl_3 \cdot CO \cdot CCl_3$
- d) CCl<sub>3</sub> · CH<sub>2</sub>OH
- 8. Which of the following compounds undergo  $E_2$  reactions more easily?
  - $(CH_3)_2 C \cdot CH_2CH_3$ a)

Br

- b)  $CH_3(CH_2)_2CH_2Cl$
- c)  $CH_3(CH_2)_2CH_2I$



- 9. Decomposition of benzene diazonium chloride by using Cu<sub>2</sub>Cl<sub>2</sub>/HCl to form chlorobenzene is
  - a) Raschig's reaction

b) Sandmeyer's reaction

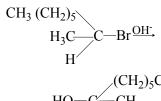
c) Kolbe's reaction

- d) Cannizaro's reaction
- 10. Isobutyl chloride and butyl chloride are:
  - a) Position isomers
- b) Chain isomers
- c) Functional isomers d) Metamers
- 11.  $CH_3Br + Nu^- \rightarrow CH_3 - Nu + Br^-$

The decreasing order of the rate of the above reaction with nucleophiles (Nu<sup>-</sup>) A to D is [Nu<sup>-</sup> = (A)PhO<sup>-</sup>, (B)AcO<sup>-</sup>, (C)HO<sup>-</sup>, (D)CH<sub>3</sub>O<sup>-</sup>]

- a) D > C > A < B b) D > C > B > A
- c) A > B > C > D
- d) B > D > C > A

12. The reaction described below is:



a)  $S_E 1$ 

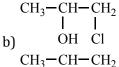
- b)  $S_N 2$
- c)  $S_N 1$

 $d)S_E2$ 

13. Identify 'Z' in the following reaction series,

$$CH_3 \cdot CH_2CH_3Br \xrightarrow{aq.NaOH} (X) \xrightarrow{Al_2O_3} (Y) \xrightarrow{HOCL} (Z)$$
:

Mixture of



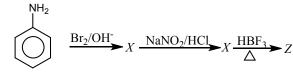
- Cl OH c)
  - CH<sub>3</sub>-CH-CH<sub>2</sub>
- Cl Cl d)
- 14. Which of the following when heated with KOH and primary amine gives carbylamine test?
  - a) CHCl<sub>3</sub>
- b) CH<sub>2</sub>Cl<sub>2</sub>
- c) CH<sub>3</sub>OH
- d) CCl<sub>4</sub>
- 15. The reagent used for dehalogenation of 1,2-dichloropropane is:
  - a) Zn dust
- b) Zn—Hg
- c) Na

d) Zn—Cu couple

- 16. CH<sub>3</sub>NH<sub>2</sub> reacts with CH<sub>3</sub>Mg*X* to give:
  - a) Acetone
- b) Alcohol
- c) Methane
- d) Ethane

- 17. Which of the following haloalkanes is most reactive?
  - a) 1-chloropropane
- b) 1-bromopropane
- c) 2-chloropropane
- d) 2-bromopropane

- 18. Iodoform is formed when ethanol is heated with:
  - a) Potassium iodide and sodium hydroxide
  - b) Iodine and aqueous potassium hydroxide
  - c) Chloroform and iodine
  - d) Iodine and potassium iodide
- 19. Tertiary alkyl halides are practically inert to  $S_N$ 2 mechanism because of:
  - a) Insolubility
- b) Instability
- c) Inductive effect
- d) Steric hinderance



The final product, is

20.

