

Topic :- Dual nature of radiation and matter

1 (c) Slope of $V_0 - v$ curve for all metals be same $\left(\frac{h}{e}\right)$, i.e., curves should be parallel

2 (b) According to de-Broglie hypothesis

$$\begin{aligned}\lambda &= \frac{h}{p} \\ &= \frac{h}{\sqrt{2mE}} = \frac{h}{\sqrt{2mqV}} \\ \therefore \lambda &= \frac{6.6 \times 10^{-34}}{\sqrt{2 \times (1.6 \times 10^{-27})(1.6 \times 10^{-19}) \times 1000}} \\ &= \frac{6.6 \times 10^{-34}}{7.16 \times 10^{-22}} \\ &= 0.9 \times 10^{-12} \text{ m}\end{aligned}$$

3 (a) $i = \frac{Ne}{t} \Rightarrow \frac{N}{t} = \frac{i}{e} = \frac{3.2 \times 10^{-3}}{1.6 \times 10^{-19}} = 2 \times 10^{16}/s$

4 (c) Speed of the cathode rays is $10^7 \text{ m/sec} - 3 \times 10^7 \text{ m/s}$

5 (b) Stopping potential $V_0 = \frac{hc}{e} \left[\frac{1}{\lambda} - \frac{1}{\lambda_0} \right]$. As λ decreases so V_0 increases

6 (b) The momentum of the photon

$$p = \frac{h}{\lambda} = \frac{hv}{c}$$

7 (c)

When pressure in a tube is reduced in the range 1 cm and 10^{-3} cm; the mean free path of moving electron in the discharge tube increases. As a result of which the electron gets higher KE while moving towards anode and then cause ionisation of the atoms with which it will collide on its ways causing excitation phenomenon.

8 (c)

Specific charge on proton $= \left(\frac{e}{m}\right)_p$

$= 9.6 \times 10^7 \text{ C} - \text{kg}^{-1}$ specific charge on α - particle,

$$\left(\frac{q}{m}\right)_\alpha = \frac{2e}{4m} = \frac{1}{2} \left(\frac{e}{m}\right)_p = \frac{1}{2} \times 9.6 \times 10^7$$

$$= 4.8 \times 10^7 \text{ C} - \text{kg}^{-1}$$

9 (c)

$$mv = \frac{h}{\lambda}$$

$$\text{or } v = \frac{h}{m\lambda} = \frac{6.6 \times 10^{-34}}{9.1 \times 10^{-31} \times 5200 \times 10^{-10}}$$

$$= 1.4 \times 10^3 \text{ ms}^{-1}$$

10 (c)

For photoelectric effect,

$$eV_0 = hv \Rightarrow v = \frac{eV_0}{h}$$

$$v = \frac{1.6 \times 10^{-19} \times 3.2}{6.6 \times 10^{-34}}$$

$$= 0.775 \times 10^{15} \text{ Hz}$$

11 (a)

On increasing wavelength of light of the photoelectric current decreases and at a certain wavelength (cut off) above which photoelectric current stops

12 (a)

If an electron and a proton propagating in the form of waves and their wavelength are same, then according to the relation

$$E = \frac{hc}{\lambda}$$

Also,

$$\lambda_{\text{electron}} = \lambda_{\text{proton}}$$

\therefore

$$E_e = E_p$$

Hence, their energies are same.

13 (b)

Stopping potential does not depend upon intensity of incident light (I)

14 (a)

Any charge in the universe is given by

$$q = ne \Rightarrow e = \frac{q}{n} \text{ (where } n \text{ is an integer)}$$

$$q_1 : q_2 : q_3 : q_4 : q_5 : q_6 :: n_1 : n_2 : n_3 : n_4 : n_5 : n_6$$

$$6.563 : 8.204 : 11.5 : 13.13 : 16.48 : 18.09$$

$$:: n_1 : n_2 : n_3 : n_4 : n_5 : n_6$$

Divide by 6.563

$$1 : 1.25 : 1.75 : 2.0 : 2.5 : 2.75 :: n_1 : n_2 : n_3 : n_4 : n_5 : n_6$$

Multiplied by 4

$$4 : 5 : 7 : 8 : 10 : 11 :: n_1 : n_2 : n_3 : n_4 : n_5 : n_6$$

$$e = \frac{q_1 + q_2 + q_3 + q_4 + q_5 + q_6}{n_1 + n_2 + n_3 + n_4 + n_5 + n_6} = \frac{73.967 \times 10^{-19}}{45}$$

$$= 1.641 \times 10^{-19} \text{C}$$

[Note : If you take 45.0743 in place of 45, you will get the exact value]

16 (b)

In the presence of inert gas photoelectrons emitted by cathode ionize the gas by collision and hence the current increases

17 (a)

$$\text{Energy of photon } E = \frac{hc}{\lambda} = mc^2;$$

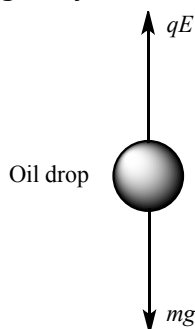
$$\text{momentum of photon } = mc = h/\lambda$$

18 (b)

In X-ray tube, target must be heavy element with high melting point

19 (c)

Robert Millikan performed the experiment to determine the charge on an electron. When a drop is suspended, its weight mg is exactly equal to the electric force applied qE , where E is electric field, q the charge, m the mass of drop and g the acceleration due to gravity.



Hence, solving for q , we get

$$q = \frac{mg}{E}$$

$$\text{Given, } m = 16 \times 10^{-6} \text{kg, } g = 10 \text{ms}^{-2},$$

$$E = 10^6 \text{V-m}^{-1}$$

$$\therefore q = \frac{16 \times 10^6 \times 10}{10^6} = 16 \times 10^{-11} \text{ C}$$

ANSWER-KEY										
Q.	1	2	3	4	5	6	7	8	9	10
A.	C	B	A	C	B	B	C	C	C	C
Q.	11	12	13	14	15	16	17	18	19	20
A.	A	A	B	A	D	B	A	B	C	C

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