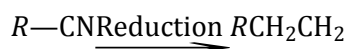


### Topic :- Amines

2 (a)



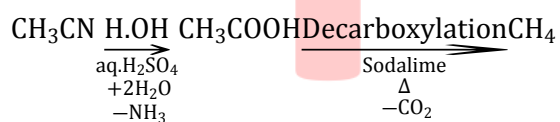
3 (d)

$C_3H_9N$  may have the structures as:  
 $(CH_3)_3N$ ,  $CH_3CH_2CH_2NH_2$ ,  $(CH_3)_2CHNH_2$   
 (a) (b) (c)

4 (b)

Tertiary amines, due to lack of H-atom, attached directly with N, does not react with benzene sulphonyl chloride ( $C_6H_5SO_2Cl$ ), i.e., Hinsberg's reagent.  $(C_2H_5)_3N$  is a tertiary amine, so does not react with  $C_6H_5SO_2Cl$ .

5 (d)

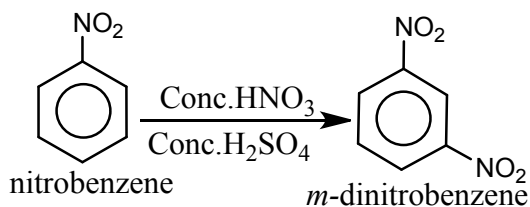


Ethane                  ethanoic acid                  methane

Nitrile                  (A)

8 (a)

Nitrobenzene on nitration gives *m*-dinitro benzene as  $-NO_2$  group is meta-directing.



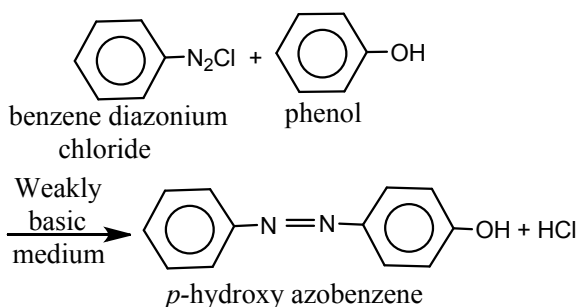
In this reaction the attacking reagent is  $NO_2^+$ .

9 (a)

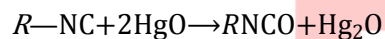
Presence of  $-\text{OCH}_3$  gp. on aniline makes it more basic than the presence of  $-\text{NO}_2$ ,  $-\text{Cl}$  or  $-\text{CH}_3$  gp.

10 (b)

Benzene diazonium chloride reacts with phenol in weakly basic medium gives *p*-hydroxy azobenzene.

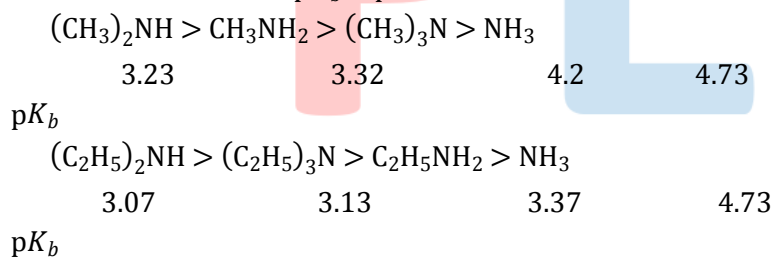


11 (c)

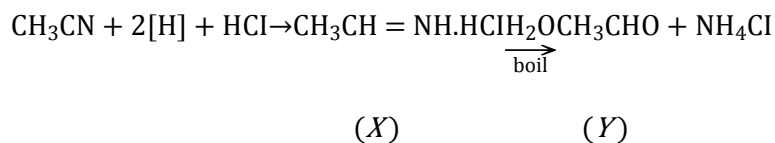
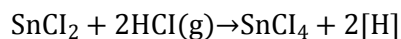


19 (d)

The abnormal trend of 3 amines is explained in terms of steric effect. Note basic order of amines on the basis of  $\text{p}K_b$  reported in Finar



20 (c)



$(X)$  is acetalimine hydrochloride and  $(Y)$  is acetaldehyde.

<b>ANSWER-KEY</b>										
<b>Q.</b>	<b>1</b>	<b>2</b>	<b>3</b>	<b>4</b>	<b>5</b>	<b>6</b>	<b>7</b>	<b>8</b>	<b>9</b>	<b>10</b>
<b>A.</b>	<b>A</b>	<b>A</b>	<b>D</b>	<b>B</b>	<b>D</b>	<b>C</b>	<b>C</b>	<b>A</b>	<b>A</b>	<b>B</b>
<b>Q.</b>	<b>11</b>	<b>12</b>	<b>13</b>	<b>14</b>	<b>15</b>	<b>16</b>	<b>17</b>	<b>18</b>	<b>19</b>	<b>20</b>
<b>A.</b>	<b>C</b>	<b>B</b>	<b>B</b>	<b>A</b>	<b>D</b>	<b>B</b>	<b>C</b>	<b>D</b>	<b>D</b>	<b>C</b>

**PE**