

Subject : CHEMISTRY DPP No. : 6 Class: XIIth

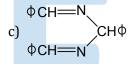
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## Topic :- Aldehydes, Ketones & Carboxylic Acids

1.	An ester ( $A$ ) with molecular formula $C_9H_{10}O_2$ was treated with excess of $CH_3MgBr$ and the complex so formed was treated with $H_2SO_4$ to give an olefin ( $B$ ). Ozonolysis of ( $B$ ) gave a ketone with molecular formul $C_8H_8O$ which shows positive iodoform test. The structure of ( $A$ ) is			
	a) $C_6H_5COOC_2H_5$		b) C <sub>6</sub> H <sub>5</sub> COOC <sub>6</sub> H <sub>5</sub>	
	c) C <sub>6</sub> H <sub>5</sub> COOCH <sub>3</sub>		d) $p$ -H <sub>3</sub> CO $-$ C <sub>6</sub> H <sub>4</sub> $-$ COCH <sub>3</sub>	
	CJ 061150000113		$u_1p_{-113}c_0 - c_{6114} - c_{0c_{113}}$	
2.	Acetone reacts with Grignard reagent to form			
	a) 3° alcohol	b) 2° alcohol	c) Ether	d) No reaction
3.	When petroleum is heated <mark>gradu</mark> ally, fi <mark>rst ba</mark> tch of vapours evolved will be rich in:			
	a) Kerosene	b) Petroleum ether	c) Diesel	d) Lubrication oil
4.	Decarboxylation of malonic <mark> acid</mark> gives			
	a) CH <sub>4</sub>	ხ) <mark>CH₃CO</mark> OH	c) Both (a) and (b)	d) None of these
5.	What is the product in the reaction			
	$CH_3CONH_2 \xrightarrow{NaOH_2/HCl} X$ ?			
	a) CH <sub>3</sub> COOH	b) CH <sub>3</sub> CONH <sub>3</sub> Cl <sup>-</sup>	c) CH <sub>3</sub> NH <sub>2</sub>	d) CH <sub>3</sub> CHO
		GH3GOWH3GI		
6.	Which of the following substances cannot be used for the replacement of —OH group in			
	organic compounds by	Cl?		
	a) S <sub>2</sub> Cl <sub>2</sub>	b) SOCl <sub>2</sub>	c) PCl <sub>3</sub>	d) PCl <sub>5</sub>
7.	Acetyl nitrate is formed when acetic anhydride reacts with			
	a) Nitrogen pentoxide	b) Nitric acid	c) Nitrous acid	d) Potassium nitrate
8.	Which one is not prepa	red from tartaric acid?		
	a) Tartar emetic	b) Fenton's reagent	c) Fehling's solution	d) Rochelle salt
9.	The reagent used in Clemmensen's reduction is			
	a) Conc. H <sub>2</sub> SO <sub>4</sub>	b) Zn-Hg /conc.HCl	c) aq. KOH	d) alc.KOH

- 10. In the reaction,  $C_6H_5COOH + CH_3 {\atop 0}H \xrightarrow{*} Ester + water$ 
  - Isotopically labeled oxygen  $(0^{18})$  is present
- b)  $0^{18}$  is present with ester
- c) 0<sup>18</sup> shifts from acid to alcohol
- d) No reaction takes place
- 11. The technique of gas chromatography is suitable for compounds which are:
  - a) Liquids
  - b) Highly volatile
  - c) Soluble in water
  - d) Vaporise without decomposition
- 12. There are several criteria of purity of organic compounds. Which is considered to be the best?
  - a) Melting point
  - b) Mixed melting point
  - c) Colour
  - d) Microscopic examination
- 13.  $\phi$ CHO + NH<sub>3</sub>→? Product is

b) 
$$\phi$$
 CH = NH



d) 
$$\phi$$
 C OH

14. The ease of hydrolysis with an alkali in the compounds

$$CH_3COCI$$
  $CH_3CO - O - COCH_3$ 

CH<sub>3</sub>COOC<sub>2</sub>H<sub>5</sub>

III

Is of the order

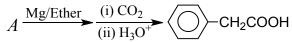
- a) I>II>III>IV
- b)IV>III>II>I
- c) I>II>IV>III
- d)II>IV>III

- 15. What is the formula of adipic acid?

  - a)  $COOH(CH_2)_4COOH$  b)  $CH_2(COOH)CH_2COOH_C$ )  $COOH(CH_2)_3COOH$
- d) None of the above
- 16. CH<sub>3</sub>CHO and C<sub>6</sub>H<sub>5</sub>CH<sub>2</sub>CHO can be distinguished chemically by:
  - a) Tollen's reagent test b) Fehling solution test c) Benedict test
- d) Iodoform test

- 17. Acrolein on complete reduction gives:
  - a) Allyl alcohol
- b) Propanol
- c) Propanal
- d) None of these

18. Identify the starting material of the following reaction



- a)  $\langle \bigcirc \rangle$  CH<sub>2</sub>CN
- b) (O)-CH<sub>3</sub>
- $_{c)}$   $\bigcirc$   $\rightarrow$   $-CH_2B$
- d) ( )-Li

- 19. Which one of the following is not a fatty acid?
  - a) Stearic acid
- b) Palmitic acid
- c) Oleic acid
- d) Phenyl acetic acid

20.  $CH_3CN \xrightarrow{H_2O} A \xrightarrow{diazomethane} B$ 

A and B are

- a) Acetamide, N-methyl acetamide
- c) Acetic acid, methyl acetate

- b) Acetic acid, ethyl ethanoate
- d) Acetamide, acetone

