

**Subject: CHEMISTRY** Class: XIIth **DPP No.: 7** Date:

Which of the following ethers form peroxide readily?

2. Association of alcohol molecules takes place because of:

a) Electrovalent bond b) Ionic bond

c) Covalent bond

d) Hydrogen bond

The reaction,  $2CH_3CH_2\dot{O}H \xrightarrow{H^+} CH_3CH_2OCH_2CH_3$  is believed to occur through the formation

Ethyl iodide on treatment with dry Ag<sub>2</sub>O will yield:

- a) Ethyl alcohol
- b) Diethyl ether
- c) Ethyl methyl ether d) Ethylene

Which of the following compounds is weakest acid?

$$O_2N$$
  $O_2N$   $O_2$   $O_3$   $O_4$   $O_4$   $O_5$   $O_$ 

d) 
$$O_2N$$
  $O_2$   $O_2$   $O_3$   $O_4$   $O_4$   $O_5$   $O_5$ 

Fusel oil is a mixture of:

- a) Alcohols
- b) Ethers
- c) Ethers and alcohols d) Alcohols and acetone

7. When benzene sulphonic acid and *p*-nitrophenol are treated with NaHCO<sub>3</sub>, the gases released respectively are

- a) SO<sub>2</sub>, NO<sub>2</sub>
- b)  $SO_2$ , NO
- c)  $SO_2$ ,  $CO_2$
- d)  $CO_2$ ,  $CO_2$

Which is correctly matched?

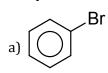
	Alcohol	α - <b>H</b>	β- <b>H</b>	Colour in Victor Meyer test
A.	X			Colourless
		3	0	
В.	Y			Blue
		1	6	
C.	Z			Red
		0	9	

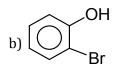
- a) A and B
- b) B and C
- c) Only C
- d) Only B
- 9. Lucas reagent is
  - a) Conc. HCI and anhydrous ZnCI<sub>2</sub>
- b) Conc. HNO<sub>3</sub> and hydrous ZnCl<sub>2</sub>
- c) Conc. HCI and hydrous ZnCI<sub>2</sub>
- d) Conc. HNO<sub>3</sub> and anhydrous ZnCl<sub>2</sub>
- 10. An aldehyde on treatment with Zn/HCl yields:
  - a) 1 alcohol
- b) 2 alcohol
- c) 3 alcohol
- d) None of these

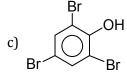
11. In the reaction,

$$A \xrightarrow{\text{K}_2\text{Cr}_2\text{O}_7} \text{acetone} \xrightarrow{\text{Oxidation}} \text{acetic acid, } A \text{ is}$$

- a) 1-propanol
- b) 2-butanol
- c) 2-propanol
- d) Ethanol
- 12. When glycerol is treated with excess of HI, it produces:
  - a) 2-iodopropane
- b) Allyl iodide
- c) Propene
- d) Glycerol tri-iodide
- 13. The product obtained by the reaction of HBr with phenol is







- d) There is no reaction
- 14. An ether is more volatile than an alcohol having the same molecular formula. This is due to
  - a) Dipolar character of ethers

- b) Alcohols having resonance structures
- c) Intermolecular hydrogen bonding in ethers d) Intermolecular hydrogen bonding in alcohols
- 15. Glycol condenses with ketones to give:
  - a) Cyclic acetals
- b) Cyclic ketals
- c) Acetaldehyde
- d) Oxalic acid

16. In the following reaction sequence

$$R - OH \xrightarrow{P + I_2} R - I \xrightarrow{AgNO_2} RNO_2 \xrightarrow{HNO_2}$$
 no reaction The alcohol is a

- a) Primary alcohol
- b) Secondary alcohol c) Tertiary alcohol
- d) Phenol

17. The explosive nitroglycerine is:

- a) A soap
- b) A salt
- c) An ester
- d) A complex compound

18. The compound CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>Br is converted into CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>OH by:

- a) Dehydration
- b) Hydrogenation
- c) Elimination
- d) Substitution

19. Consider the following reaction,

ethanol 
$$\xrightarrow{PBr_3} X \xrightarrow{alc. KOH}$$

$$Y \xrightarrow{\text{(i) H}_2\text{SO}_4 \text{ at room temperature}} Z;$$

The product *Z* is:

- a) CH<sub>3</sub>CH<sub>2</sub>OH
- b)  $CH_2 = CH_2$
- c)  $CH_3CH_2$ —O— $CH_2$ — $CH_3$
- d)  $CH_3 CH_2 O SO_3H$



20. Glycerol reacts with potassium bisulphate to produce

- a) Allyl iodide
- b) Allyl sulphate
- c) Acryl aldehyde
- d) Glycerol trisulphate