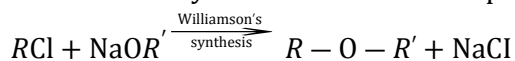


### Topic :- Alcohols, Phenols & Ethers

- 3 (c) Williamson's synthesis is used for the preparation of ethers.



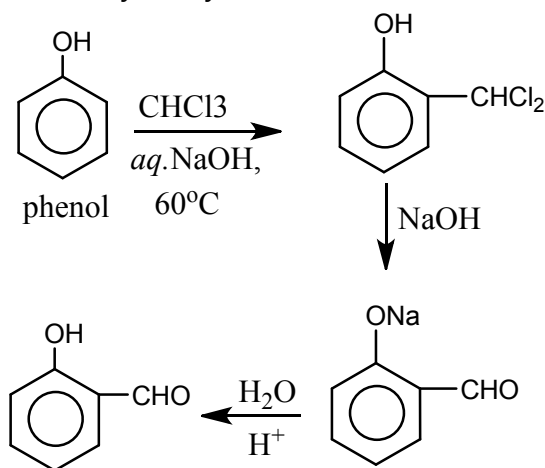
ether

- 4 (b) Starch  $\xrightarrow{\text{Diastase}}$  Maltose  $\xrightarrow{\text{Maltase}}$  Glucose  $\xrightarrow{\text{Zymase}}$  Alcohol

- 5 (a) Destructive distillation of wood gives Pyroligneous acid from which CH<sub>3</sub>OH is obtained by fractional distillation.

- 6 (c)  $-\text{COOH} \xrightarrow{\text{LiAlH}_4} -\text{CH}_2\text{OH}$

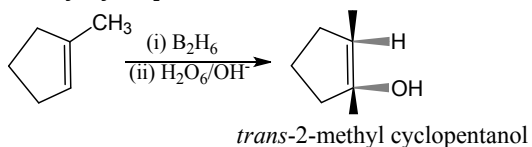
- 7 (b) Reimer-Tiemann Reaction In this reaction phenol reacts with chloroform and alkali to form salicylaldehyde.



salicylaldehyde

- 8 (d) During hydroboration-oxidation, addition of H<sub>2</sub>O across the double bond occurs anti to Markownikoff's rule and since the stereochemistry of addition *cis*, therefore *trans*-2-

methylcyclopentanol is formed



9

**(c)**

CH<sub>3</sub>OH is carbinol; CH<sub>3</sub>CH<sub>2</sub>OH is methyl carbinol and so on.

11

**(c)**

Both possess antiseptic nature.

12

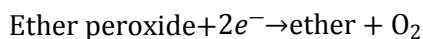
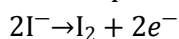
**(a)**

The percentage of alcohol is expressed as proof spirit for tax lavy. It contains 57.1 % (by vol.) or 48% (by wt.) of alcohol.

13

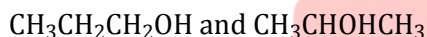
**(a)**

Ether peroxide oxidises KI into I<sub>2</sub> and itself gets reduced to ether. Therefore, KI is added to remove peroxides from ethers.



14

**(a)**



15

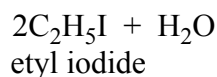
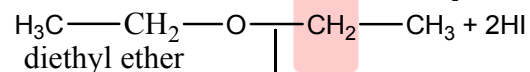
**(d)**

Lower members are soluble in water due to H-bonding and solubility decreases with increasing hydrophobic character.

18

**(c)**

Ether on reaction with excess of HI produce two molecules of alkyl halide.



Ethyl iodine

When equimolar quantities of ether and HI are present, then one molecule of alkyl halide and one molecule of alcohol are formed.

19

**(a)**

It is a substitute of petrol.

<b>ANSWER-KEY</b>										
<b>Q.</b>	<b>1</b>	<b>2</b>	<b>3</b>	<b>4</b>	<b>5</b>	<b>6</b>	<b>7</b>	<b>8</b>	<b>9</b>	<b>10</b>
<b>A.</b>	<b>A</b>	<b>A</b>	<b>C</b>	<b>D</b>	<b>A</b>	<b>C</b>	<b>B</b>	<b>D</b>	<b>C</b>	<b>A</b>
<b>Q.</b>	<b>11</b>	<b>12</b>	<b>13</b>	<b>14</b>	<b>15</b>	<b>16</b>	<b>17</b>	<b>18</b>	<b>19</b>	<b>20</b>
<b>A.</b>	<b>C</b>	<b>A</b>	<b>A</b>	<b>A</b>	<b>D</b>	<b>D</b>	<b>B</b>	<b>C</b>	<b>A</b>	<b>D</b>

**PE**